

# Biomedical Applications of Single-Walled Carbon Nanotubes: Toward Design of Novel Optical Sensors

Bin Mu, Ph.D.

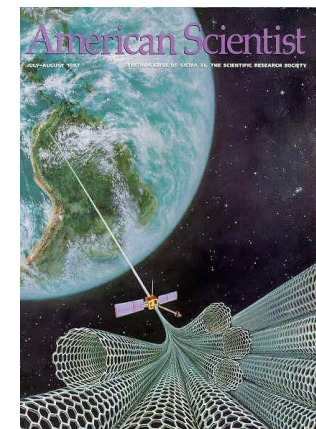
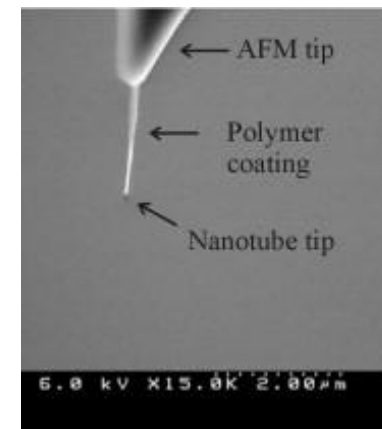
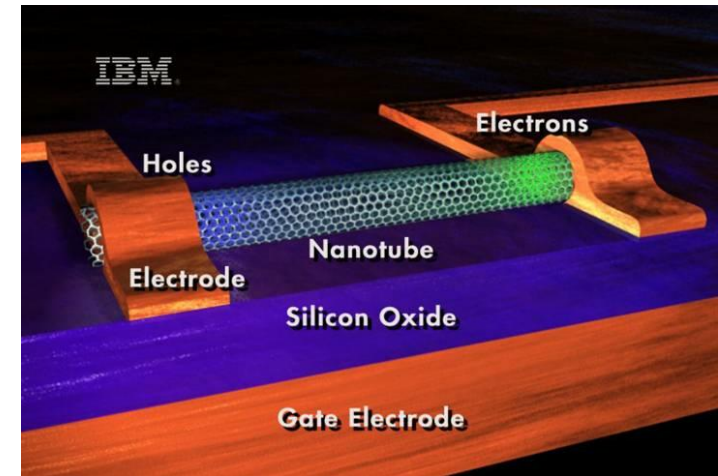
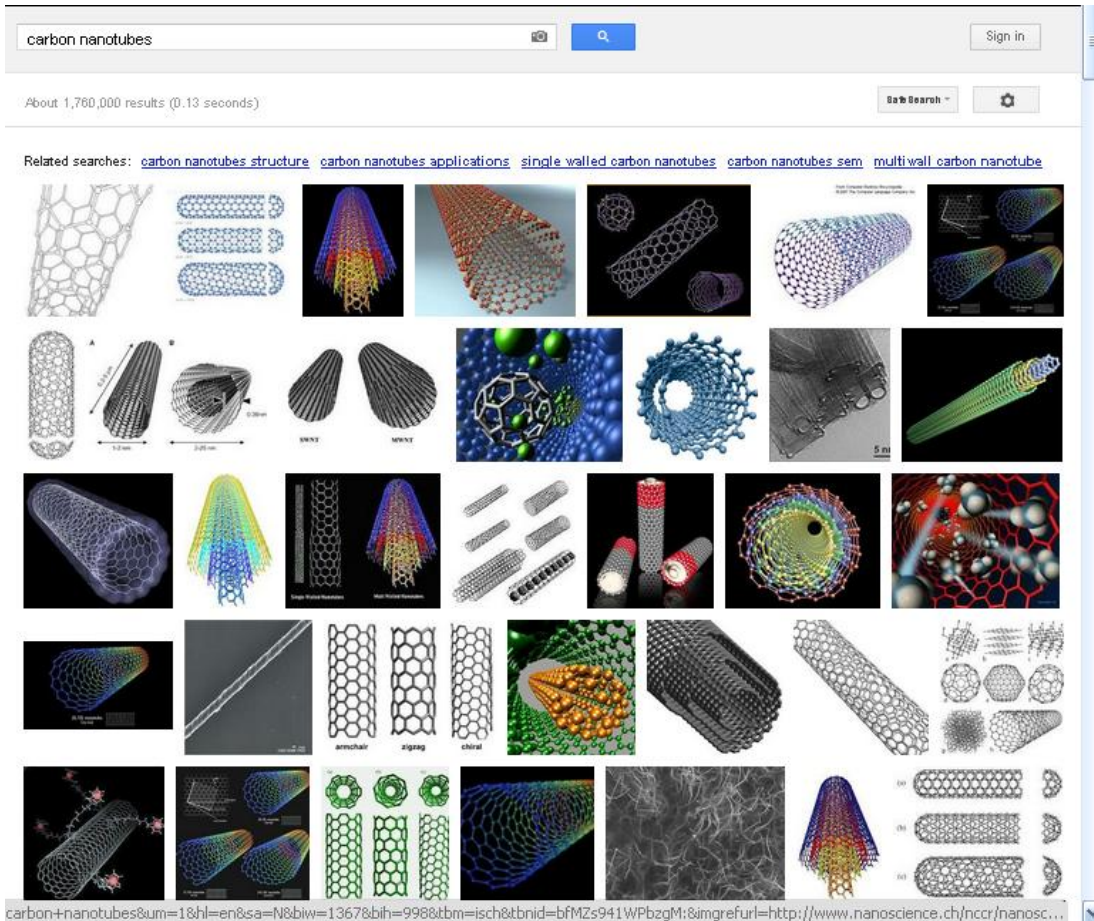
Prof. Michael S. Strano

Department of Chemical Engineering  
Massachusetts Institute of Technology

2012/06/26

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# Carbon Nanotubes are Really Cool!



[http://www.research.ibm.com/nanoscience/publications\\_optics.html](http://www.research.ibm.com/nanoscience/publications_optics.html)  
[http://www.nanotechbuzz.com/50226711/carbon\\_nanotube\\_afm\\_tip.php](http://www.nanotechbuzz.com/50226711/carbon_nanotube_afm_tip.php)

# Outline

Introduction of Single-Walled Carbon Nanotubes (SWNTs)  **What ?**

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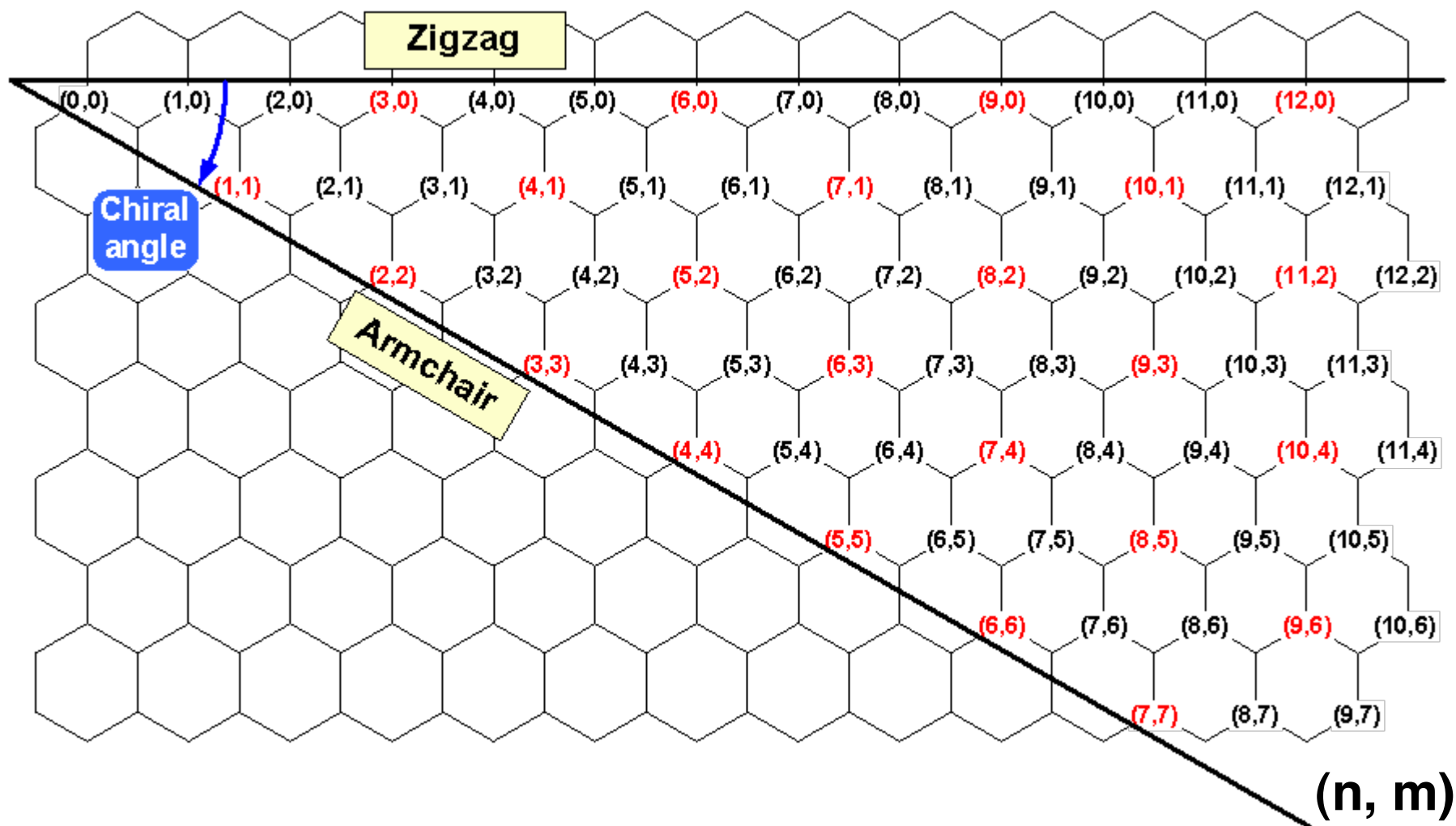
Biomedical Applications of SWNTs  **Where ?**

- Nitric Oxide Recognition
- Hydrogen Peroxide Measurements
- Detection of DNA Configuration
- Glucose Sensor

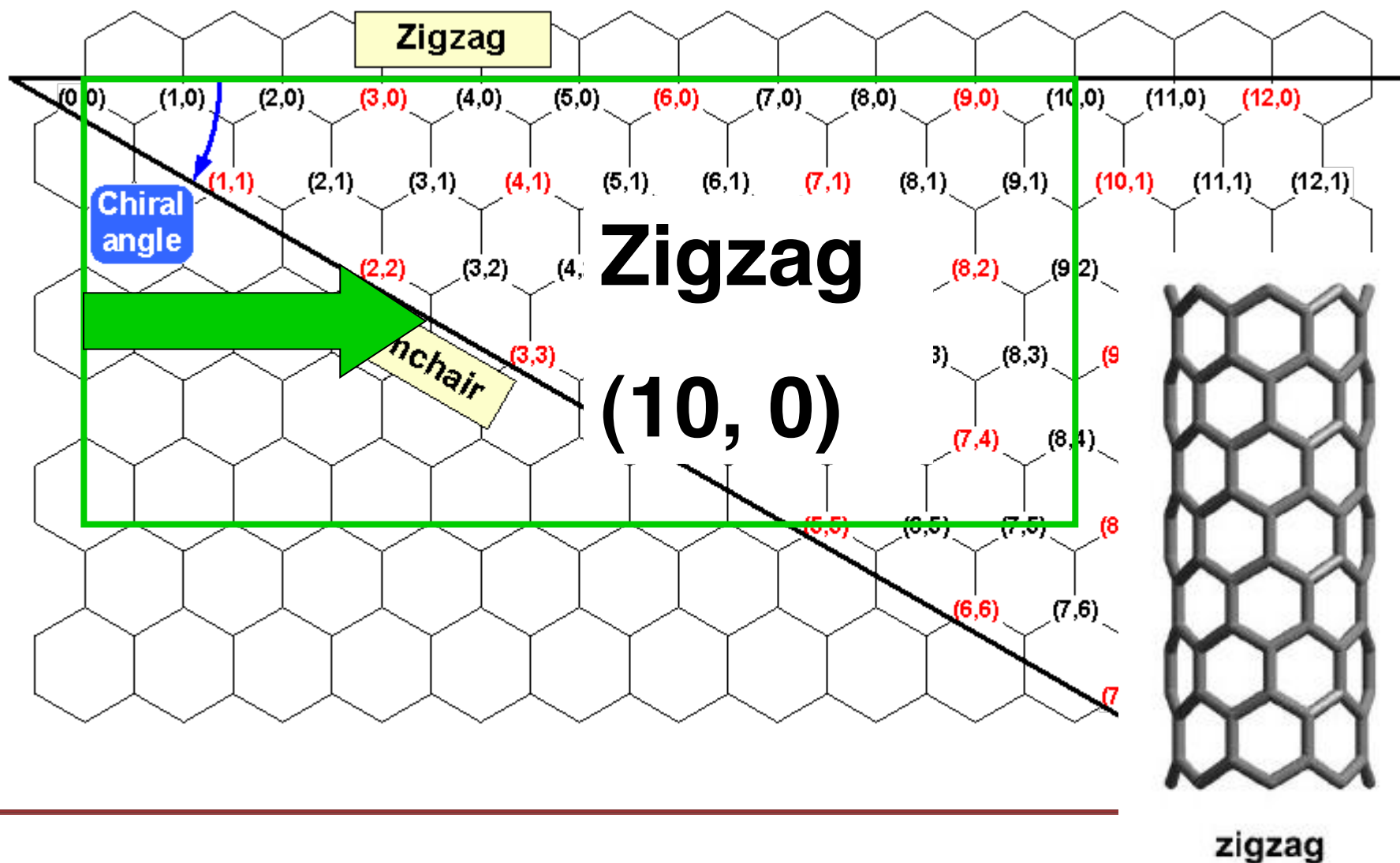
Rationale of Molecular Recognition  **Why ?**

Conclusions

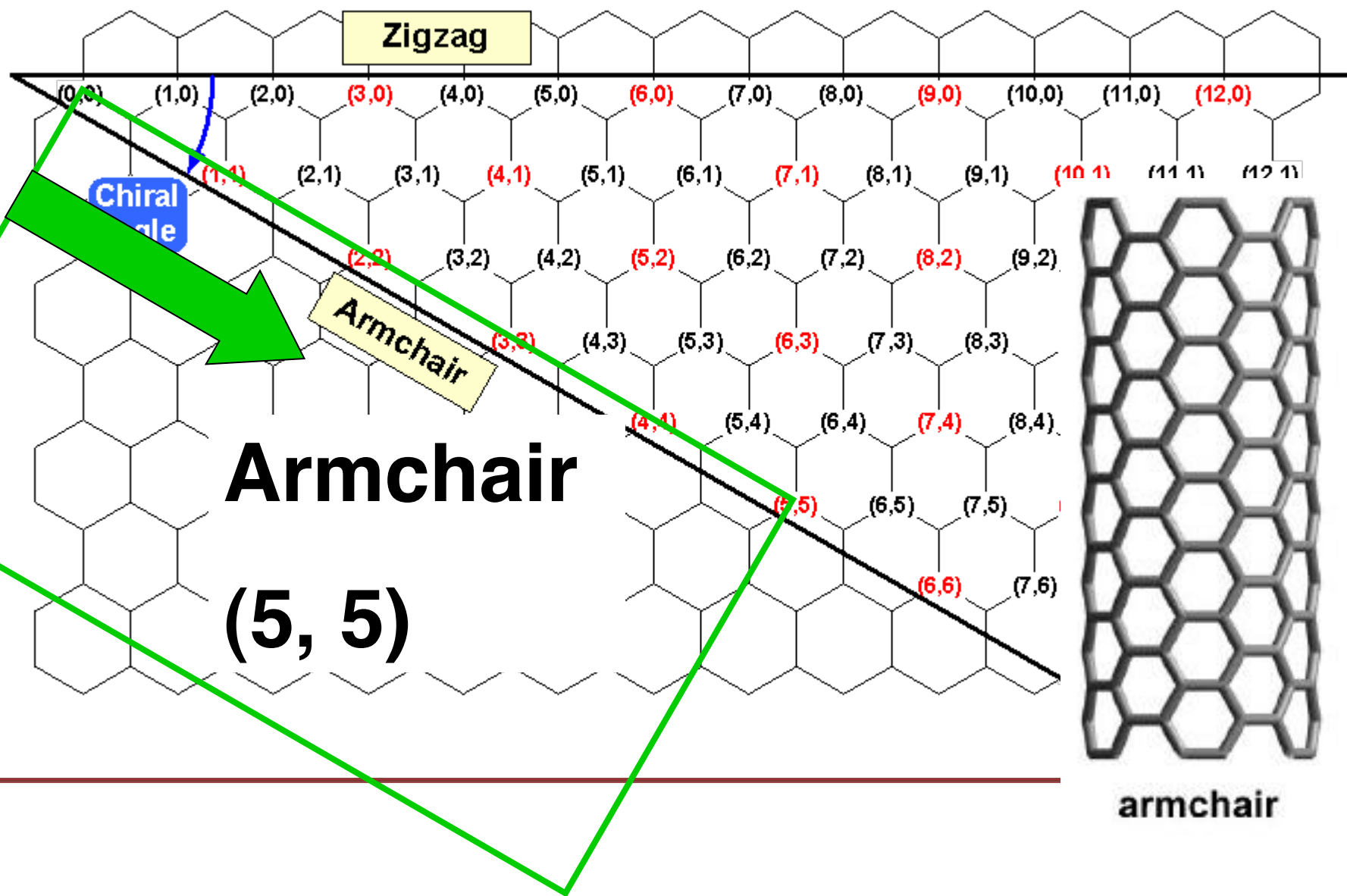
# Construction of SWNTs from a Graphene Sheet



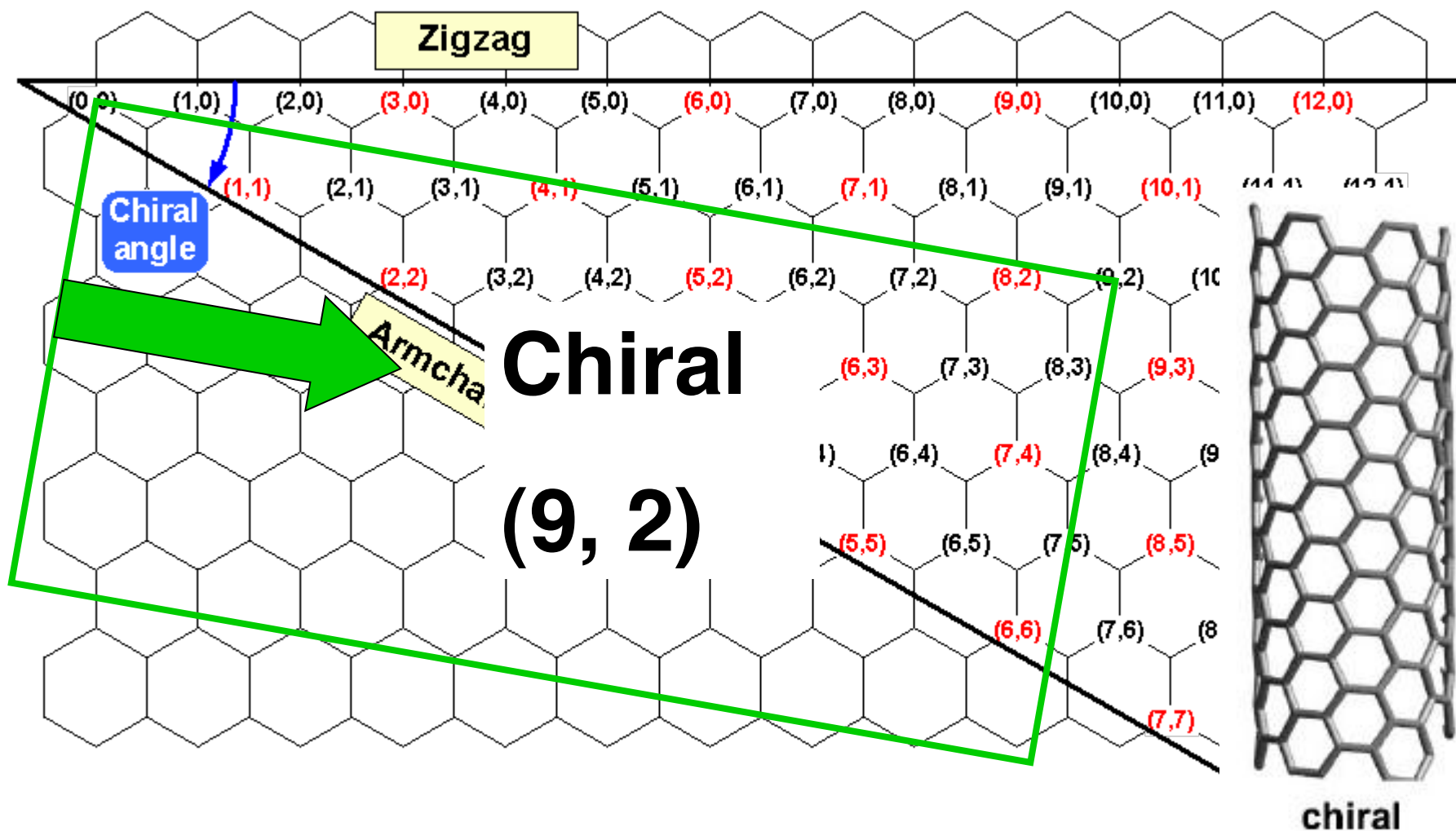
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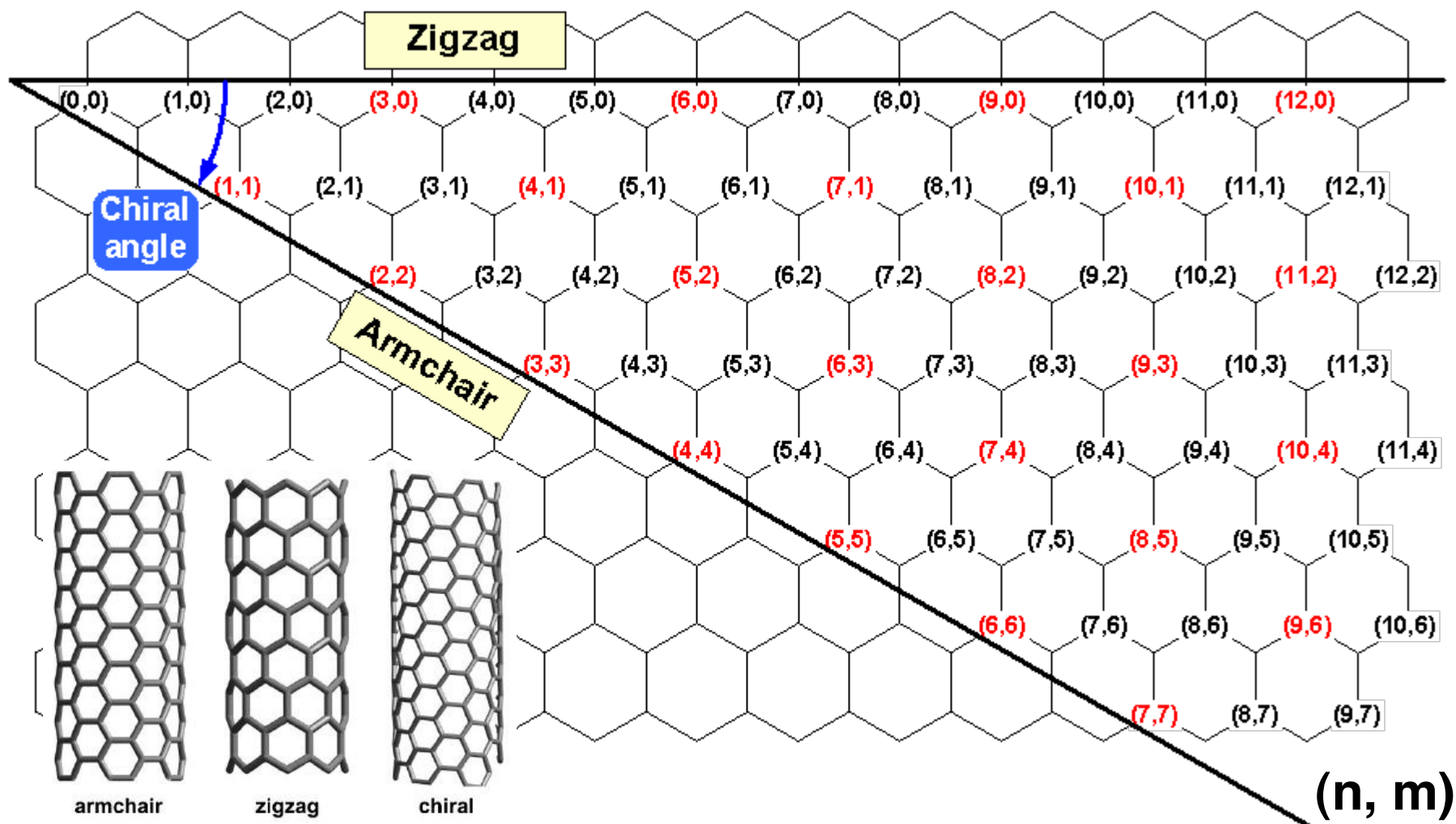
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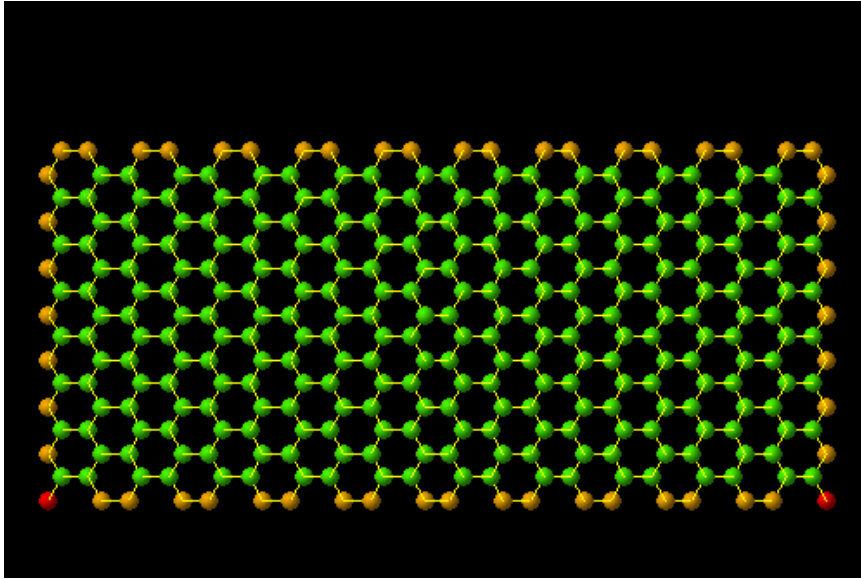
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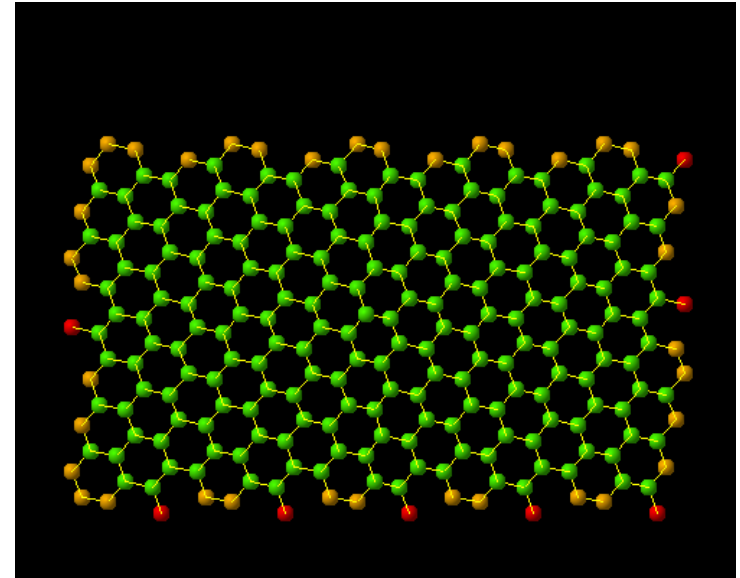
# Metallic SWNT vs. Semiconducting SWNT



$$(n, m) = (10, 10)$$

$$\text{mod}\left(\frac{n - m}{3}\right) = 0$$

Metallic SWNT



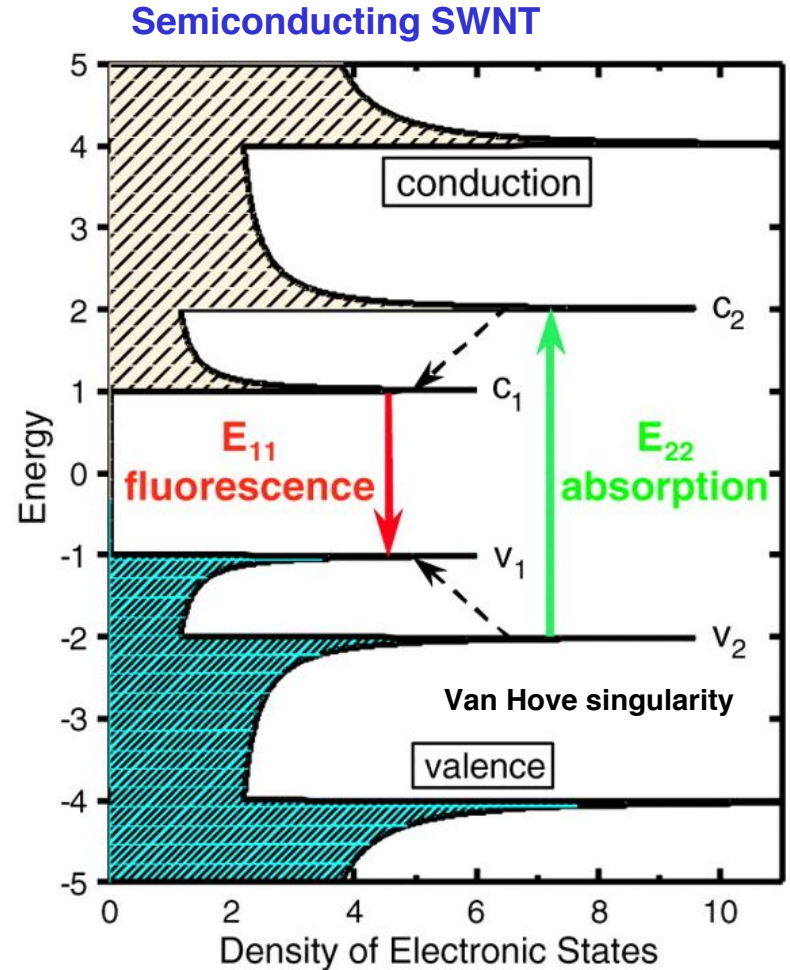
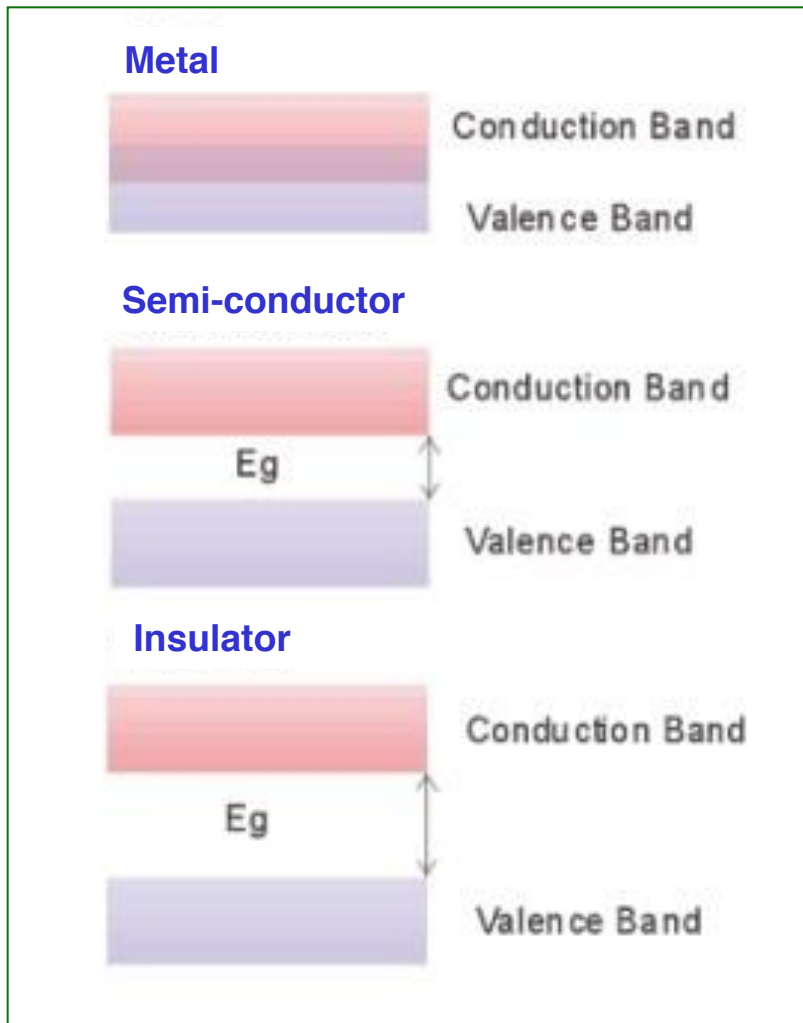
$$(n, m) = (10, 5)$$

$$\text{mod}\left(\frac{n - m}{3}\right) = 2 \neq 0$$

Semiconducting SWNT

<http://www.photon.t.u-tokyo.ac.jp/>

# Band Gap Theory



[http://people.seas.harvard.edu/~jones/es154/lectures/lecture\\_2/energy\\_gap/energy\\_gap.html](http://people.seas.harvard.edu/~jones/es154/lectures/lecture_2/energy_gap/energy_gap.html)

Sergei M. Bachilo, etc. *Science*, 2002, 298, 2361

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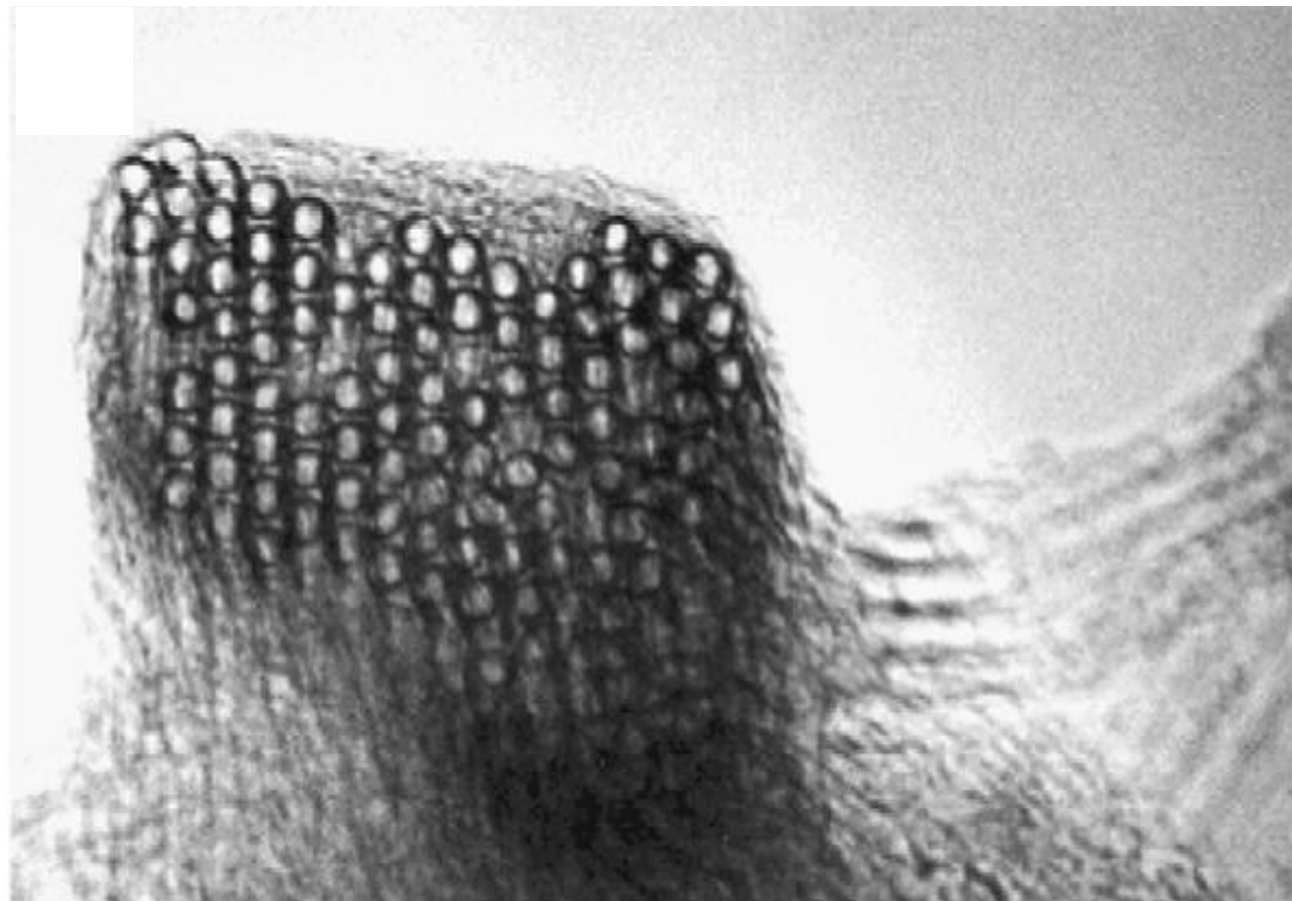
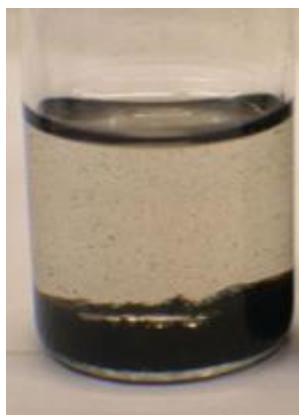
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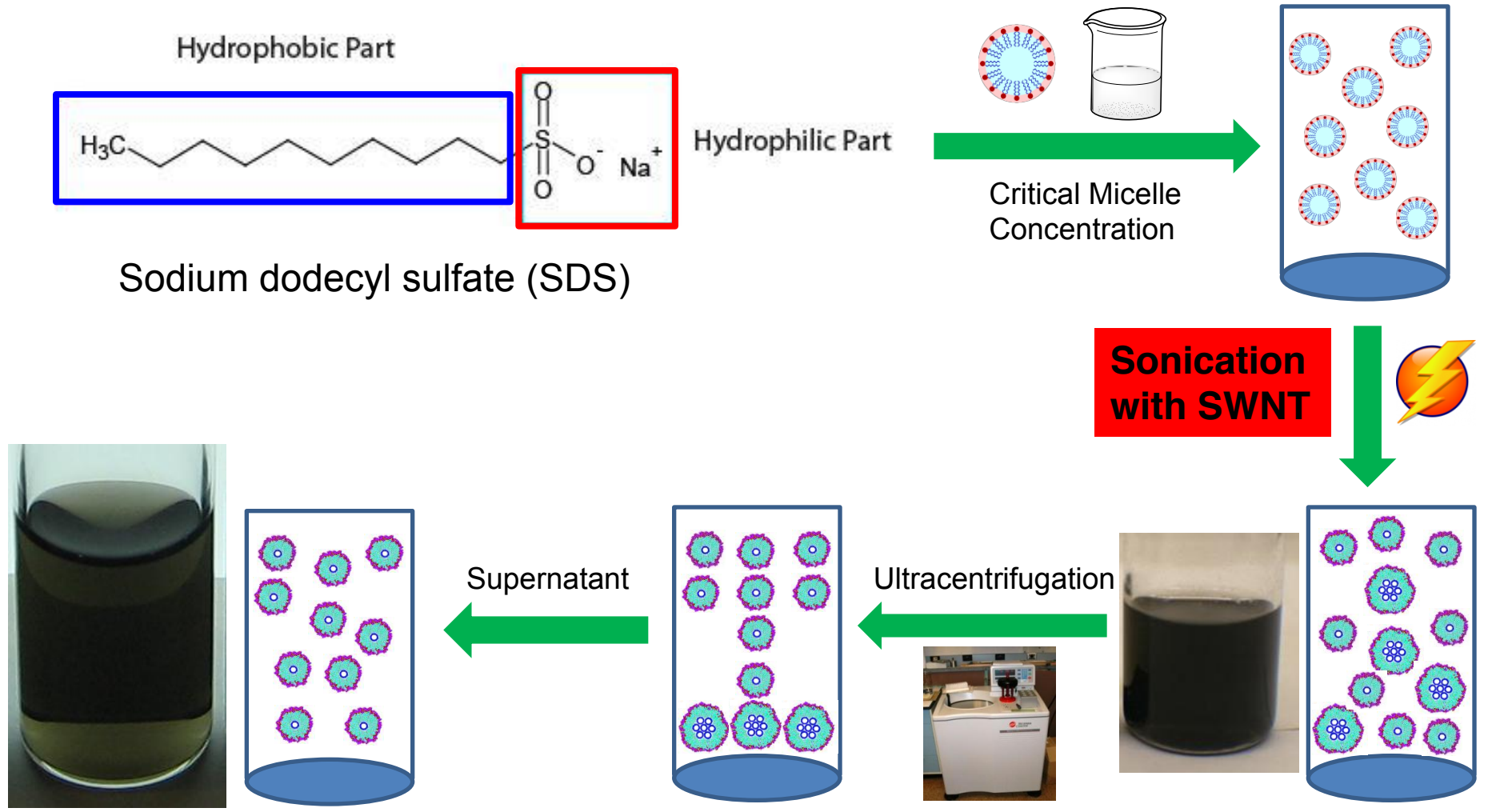
# As-Synthesized Bundled SWNTs



<http://maju-indonesia-ku.forumotions.com/t31-bahan-dasar-pembuatan-wahana-siluman>

*Advances in Colloid and Interface Science* 128–130  
(2006) 37–46

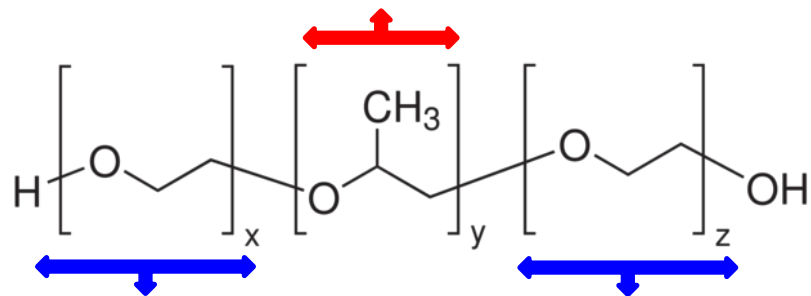
# SWNT Solubilization in Water by Surfactants



1. <http://en.wikipedia.org/wiki/File:Micelle.png>

# SWNT Solubilization in Water by Polymers

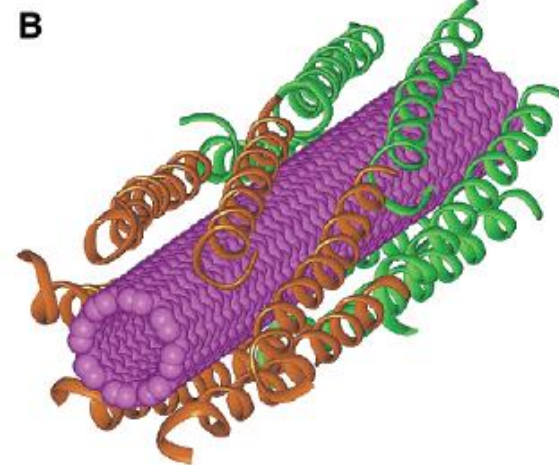
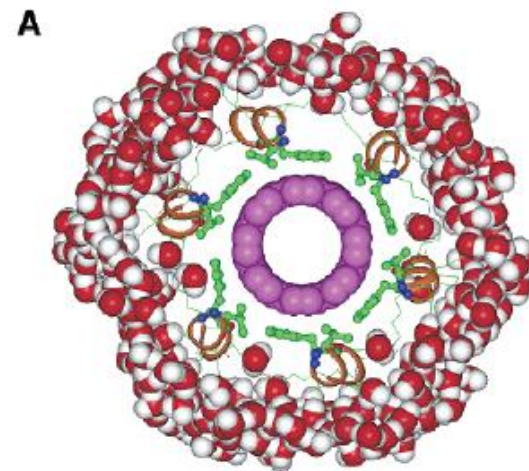
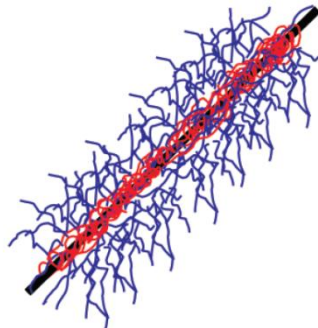
Hydrophobic part



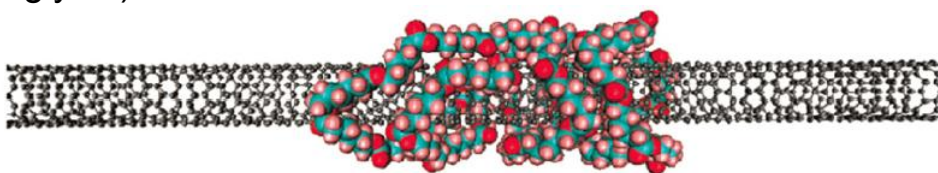
Hydrophilic part

Hydrophilic part

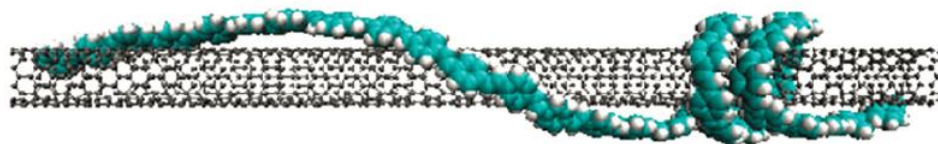
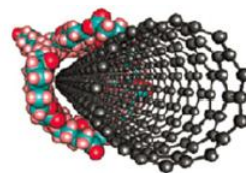
Poly(ethylene glycol)-block-poly(propylene glycol)-block-poly(ethylene glycol): **PEG-PPG-PEG**



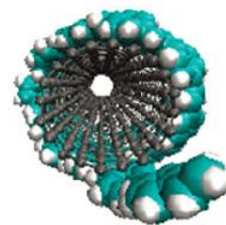
Peptides-SWNTs



(a)

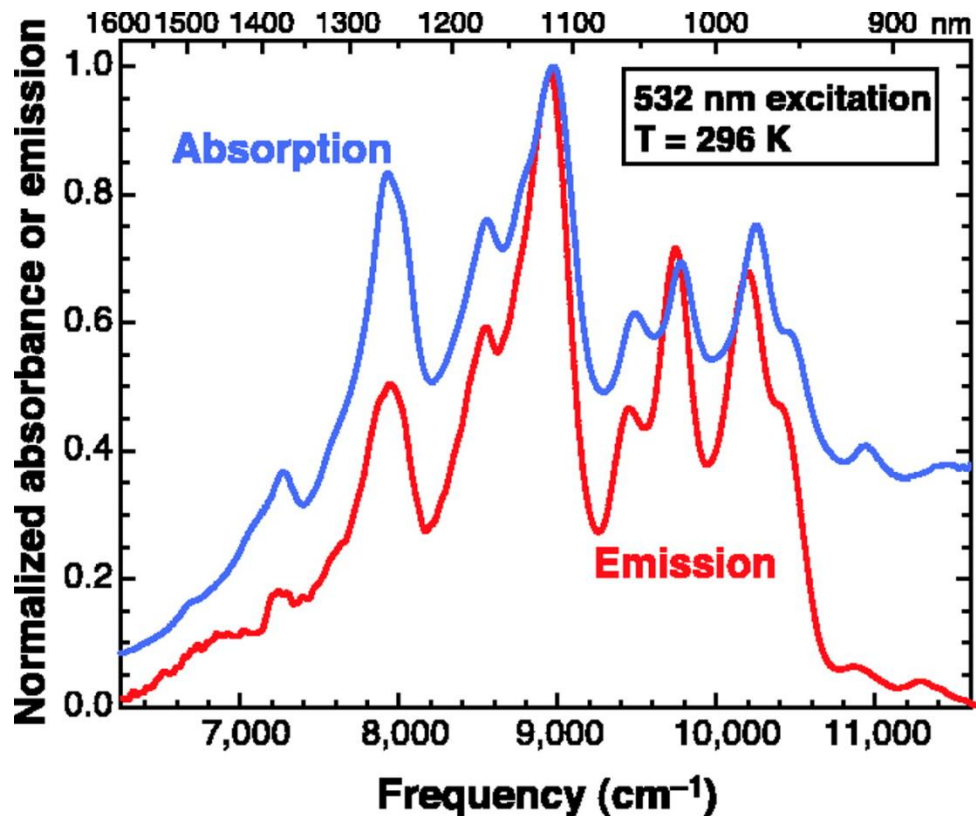
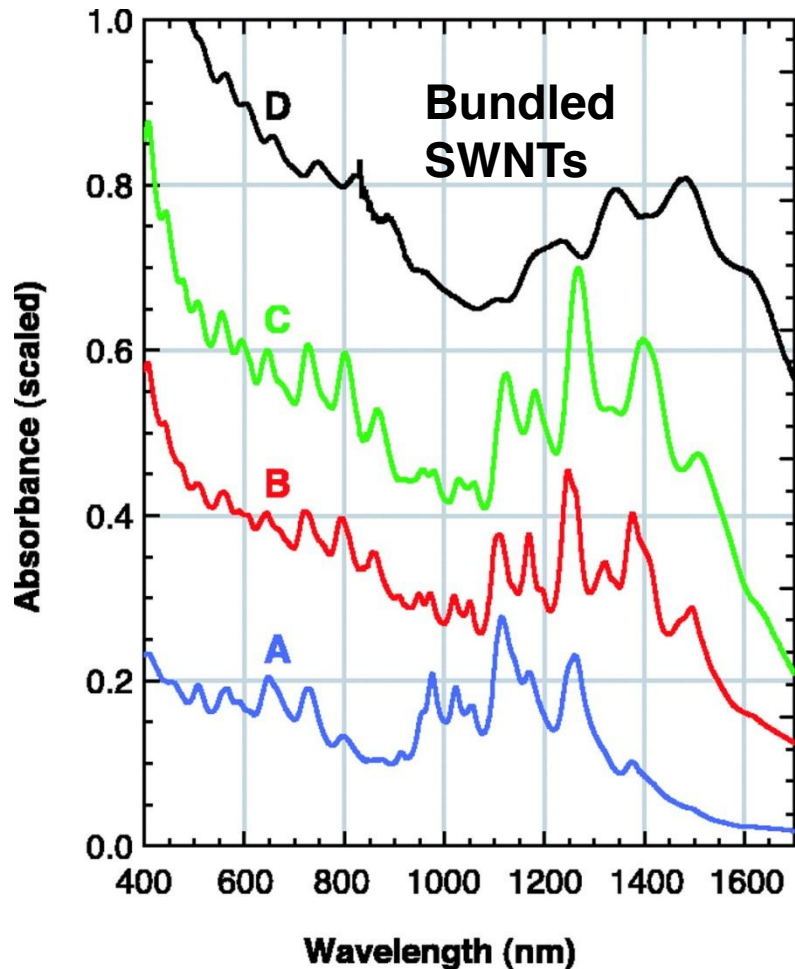


(b)



*Langmuir* 2008, 24, 4625-4632; *JACS*, 2003, 125, 1770 ;  
*Composites Science and Technology*, 2011, 72, 72

# UV-Visible Absorption and Fluorescence Emission Spectra of SWNTs



Consistent peak position from absorption and emission indicates that the major components are individual SWNTs.

*Science* 26 July 2002: Vol. 297 no. 5581 pp. 593-596

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# Advantages of Using SWNTs for Optical Sensing

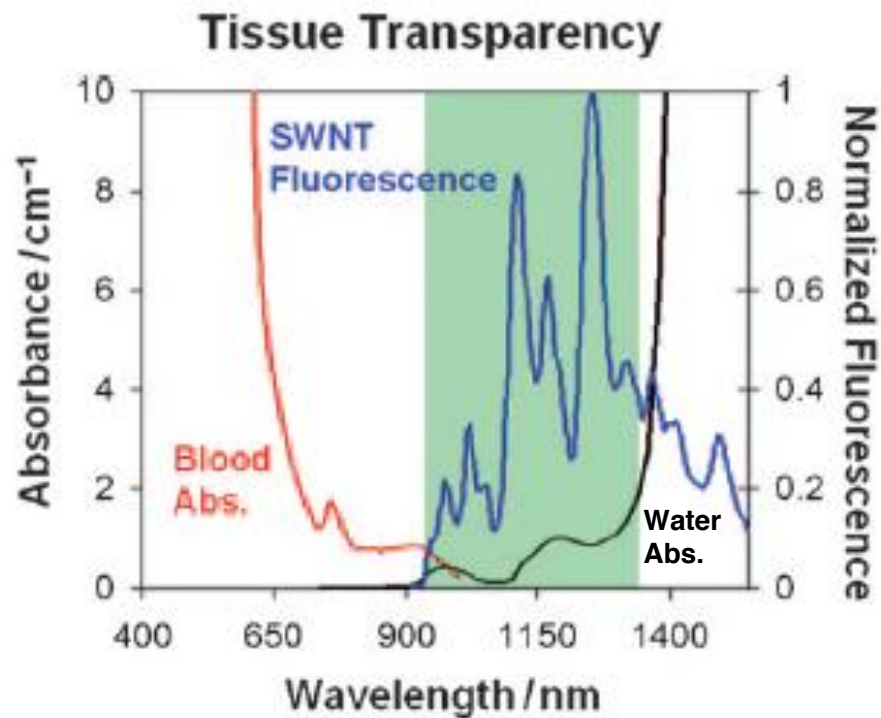
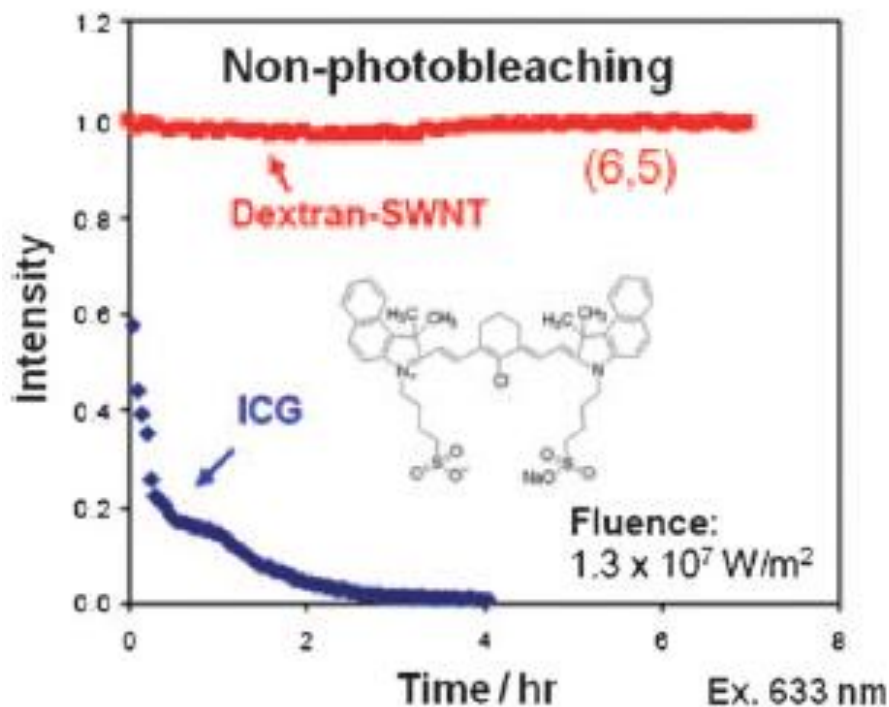


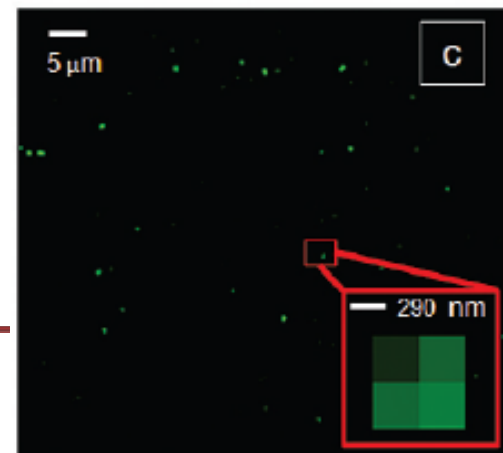
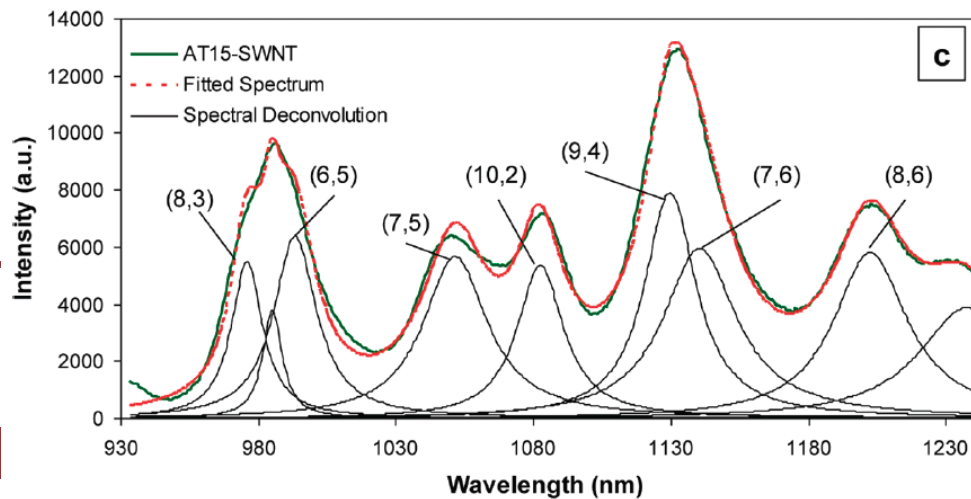
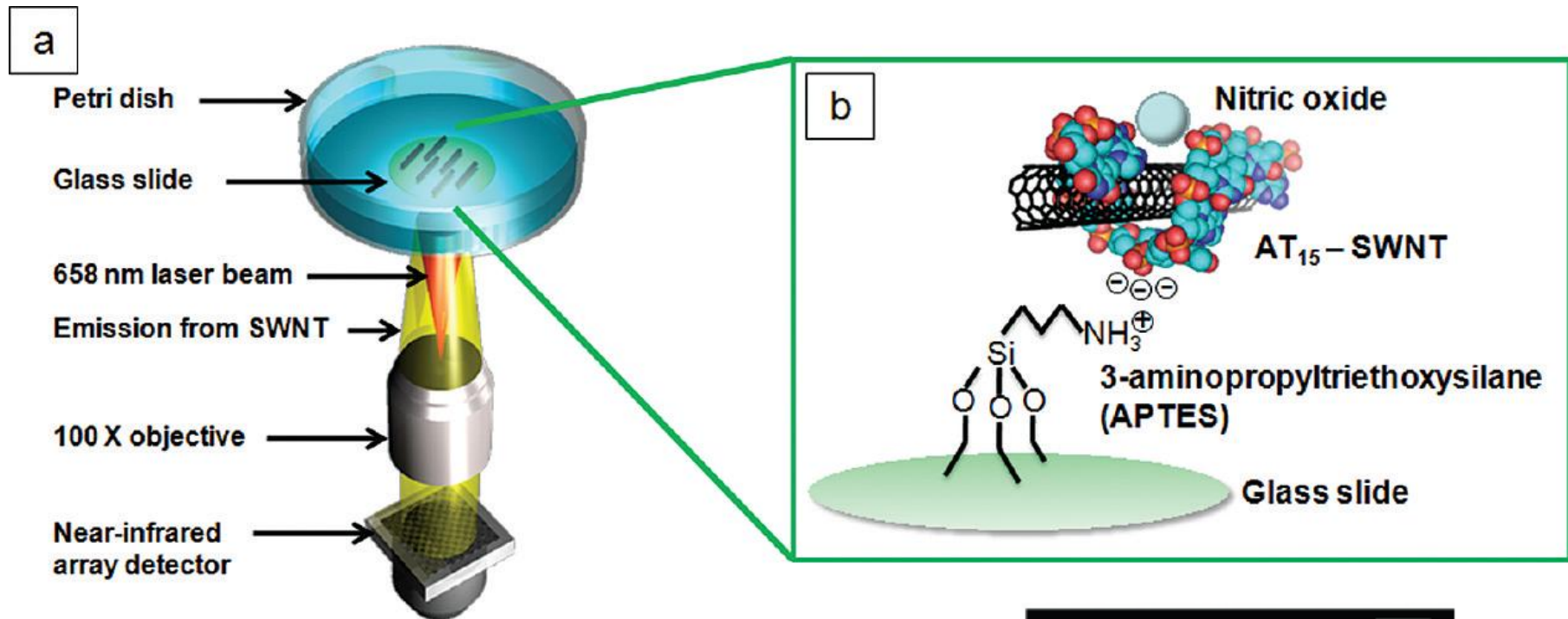
Table 1. Photobleaching tendency of common organic and nanoparticle fluorophores. Adapted from Ref. [19].

Material	Photobleaching rate constant [ $\text{h}^{-1}$ ]	Fluence [ $\text{mWcm}^{-2}$ ]	Nominal sensor lifetime
IR-Dye 78-CA <sup>[33]</sup>	250.92	600	3.2 s
Cy5 <sup>[33]</sup>	20.52	600	39.1 s
Indocyanine Green <sup>[32]</sup>	0.0412	28	5.4 h
Type II NIR QD <sup>[33]</sup>	0.0827	600	2.7 h
SWCNT <sup>[1, 22g]</sup>	0	$1.0 \times 10^6$	$\infty$

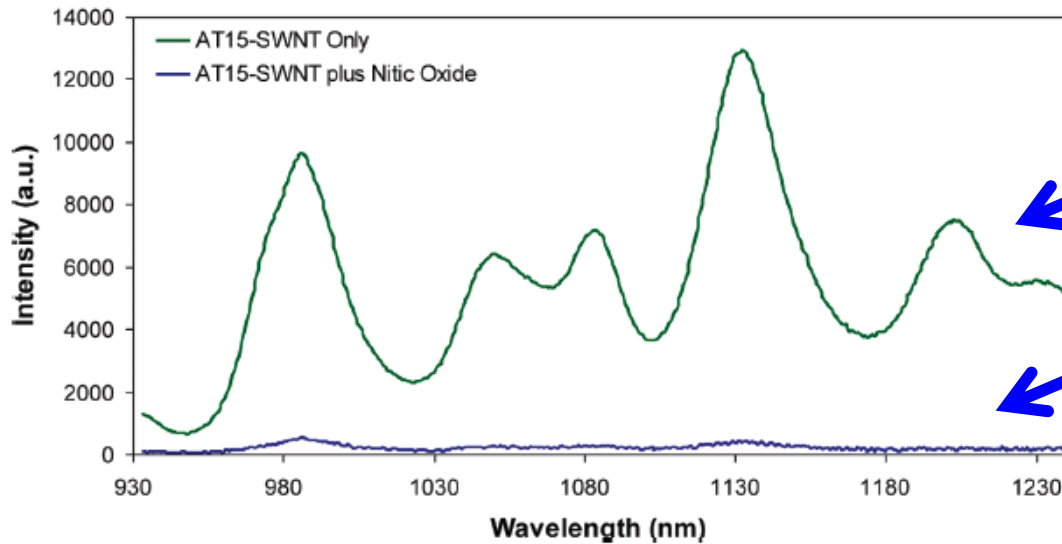
That is what we want!

*ChemSusChem*, 2011, 4, 848

# nIR Fluorescence Sensor Based on SWNTs

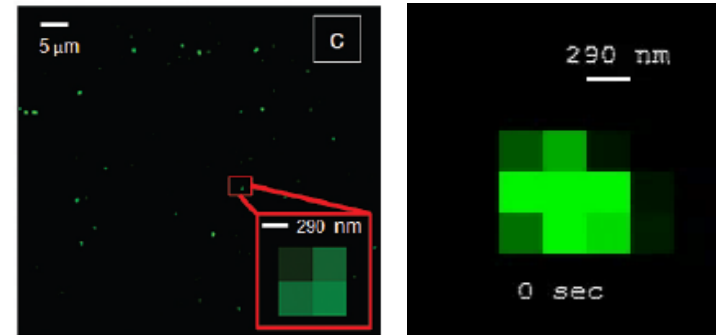
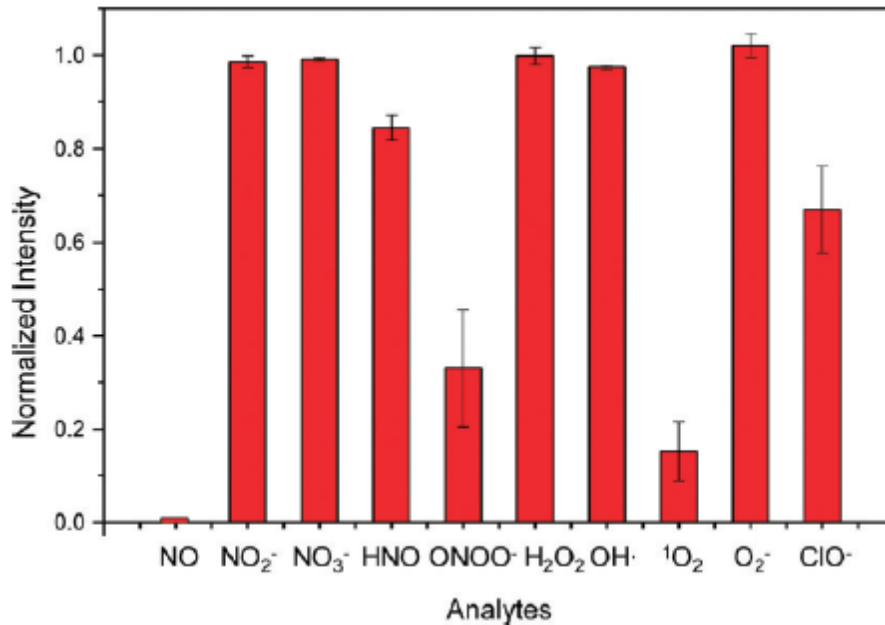


# Nitric Oxide Recognition



Near Infrared fluorescence spectra of SWNTs dispersed using AT15

After adding NO, fluorescence got quenched!

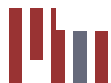
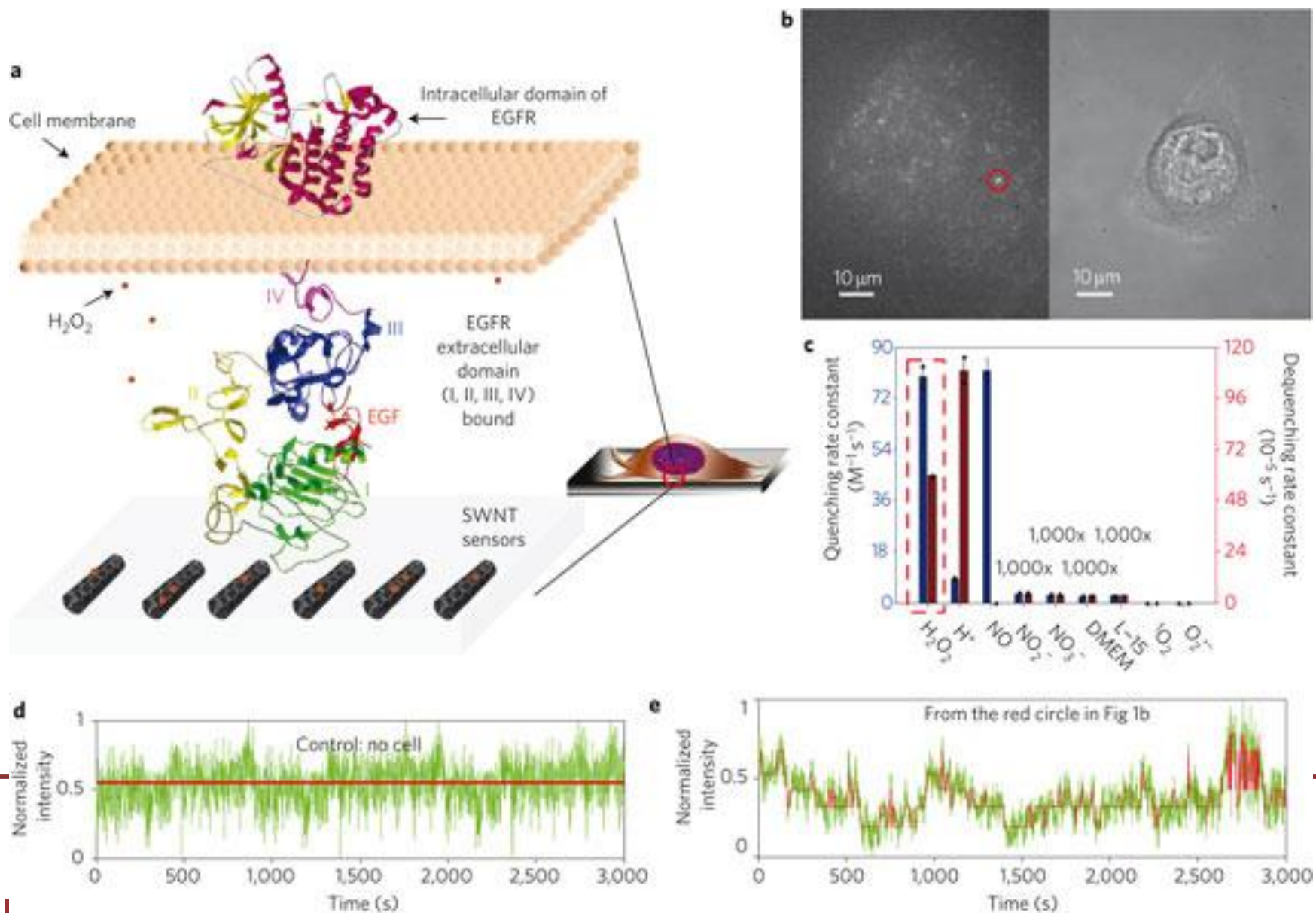


Selectivity measurement demonstrated that NO is the best quencher!

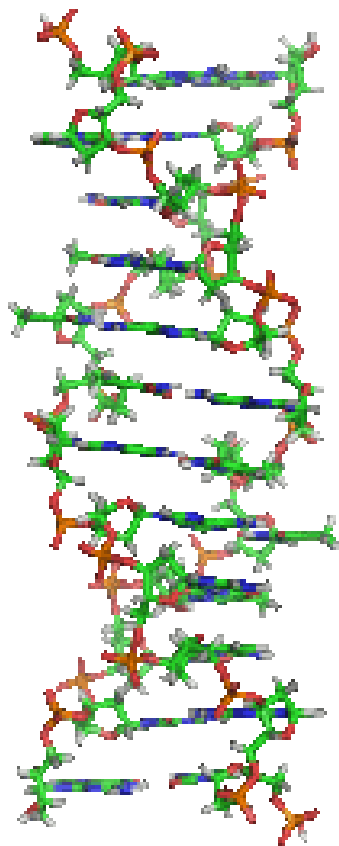
*JACS*, 2011, 133, 567



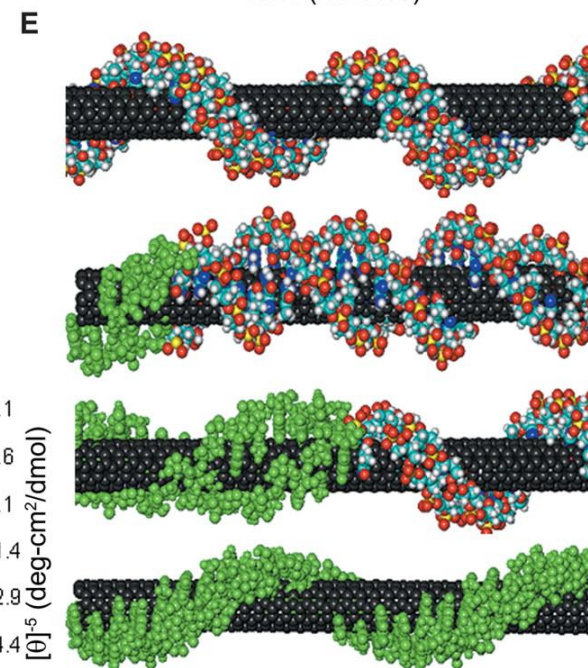
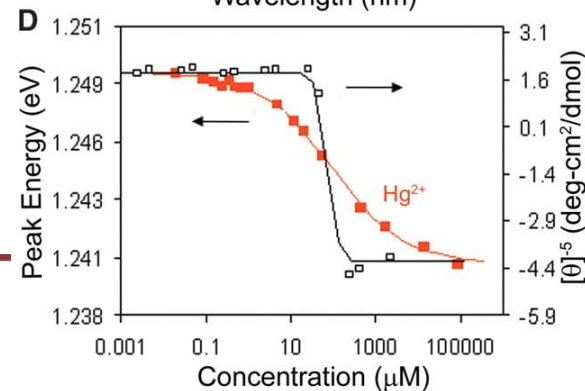
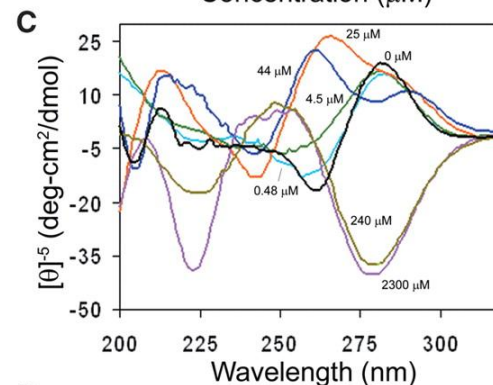
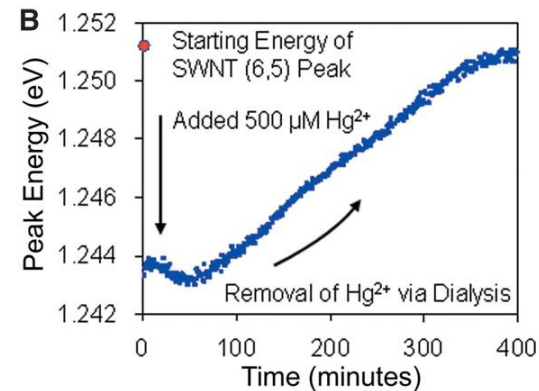
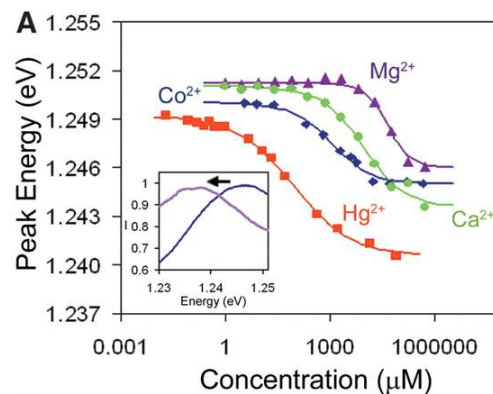
# Hydrogen Peroxide Detection



# Detection of DNA Configuration



[http://en.wikipedia.org/wiki/File:Z-DNA\\_orbit\\_animated\\_small.gif](http://en.wikipedia.org/wiki/File:Z-DNA_orbit_animated_small.gif)



# Opportunity for Development of Glucose Sensor

- **Diabetes Mellitus**

- Affects *17.9 million people* in U.S (5-10% Type I).
- *1.6 million new cases* are diagnosed each year
- *7<sup>th</sup> leading cause of death* in 2010

- **Need**

- Simple and reliable sensor for *real-time glucose detection* (< 10 min per measurement)
  - Help patients meet *targets in blood glucose* to reduce complications
-

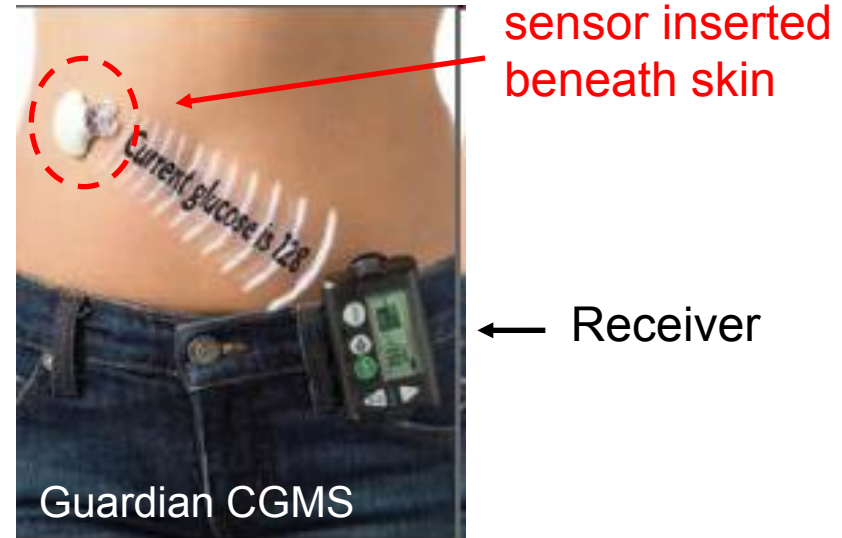
# Current State of the Art

## Commercially Available Devices

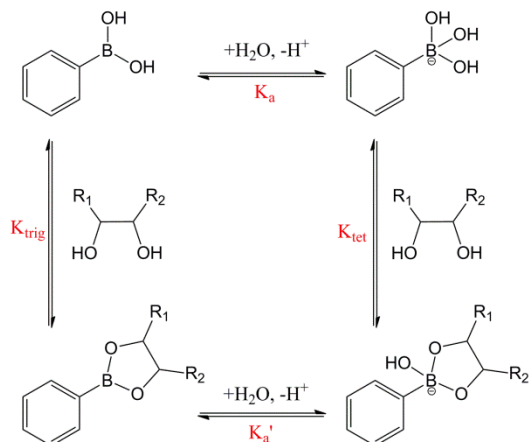
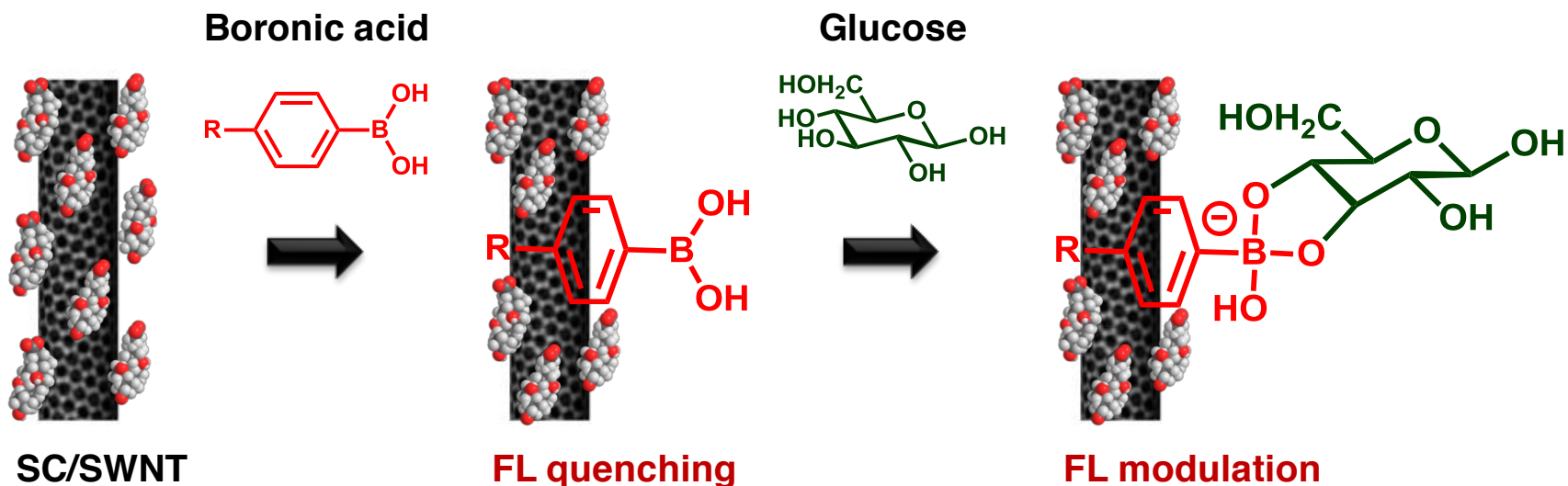
- MiniMed Guardian CGMS
- Dexcom SEVEN CGM
- Abbott FreeStyle Navigator

## Disadvantages

- Sensor lifetime (3-7 days)
- Frequent calibration
- Open wound
- More susceptible to biofouling



# Boronic Acid Based Sensor for Glucose Detection

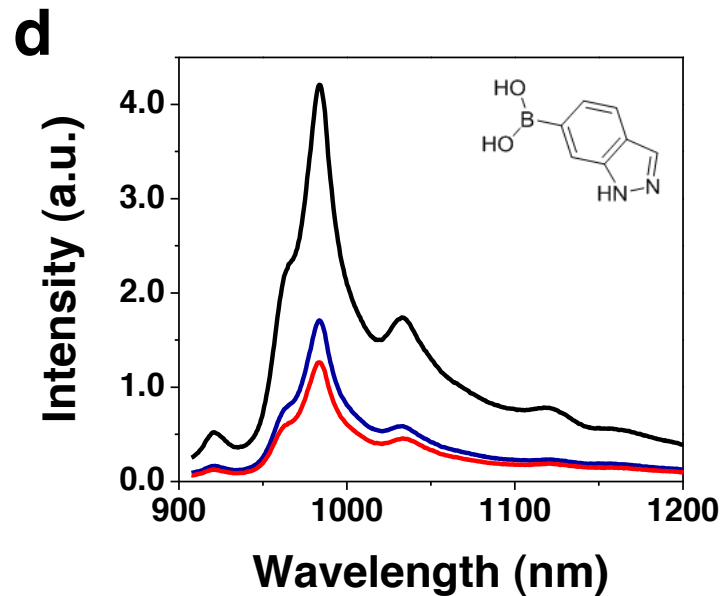
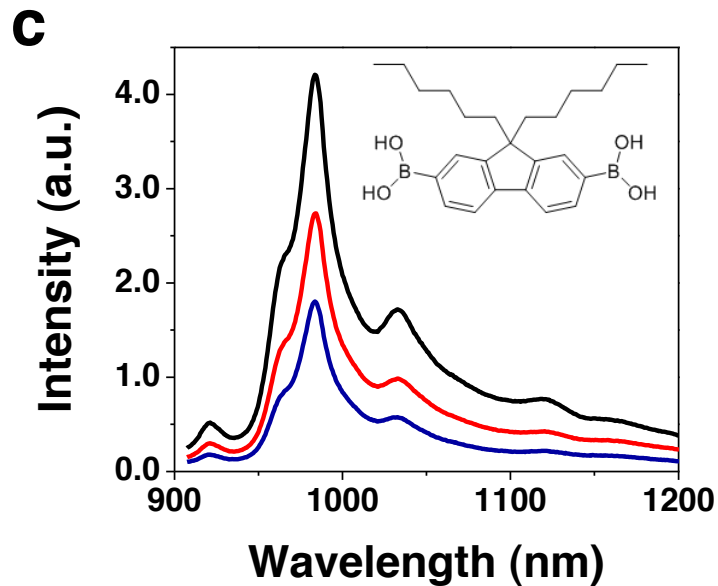
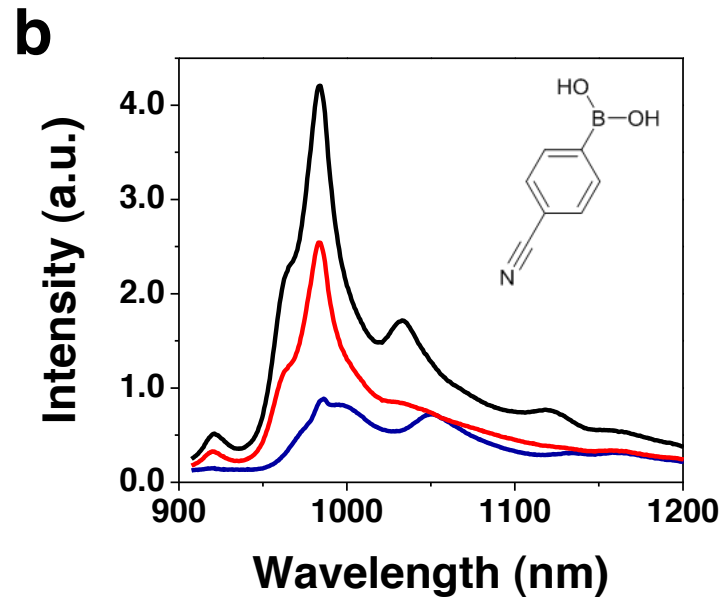
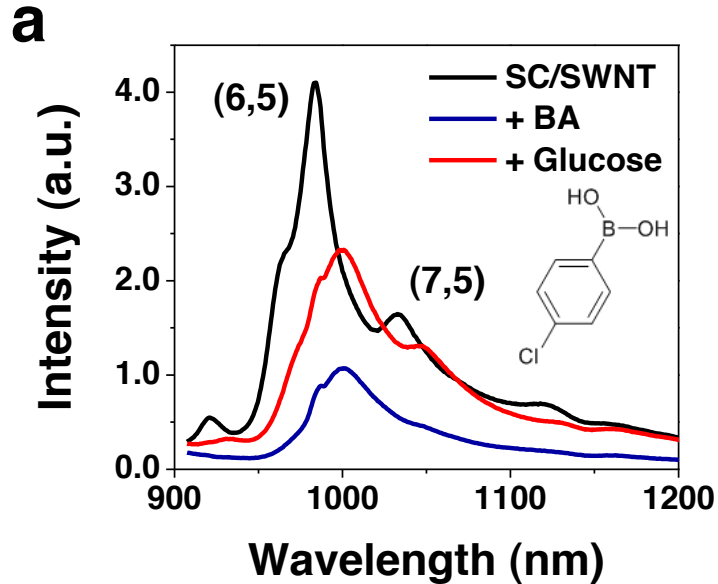


**Basic Mechanism: Transduction of glucose binding to boronic acid on SWNT to SWNT fluorescence**

**Two Fluorescence Detection Signals:**  
**Wavelength Shift ( $\Delta\lambda$ )**  
**Intensity Change ( $\Delta I$ )**

*ACS Nano*, 2012, 6, 819

# Representative Responses



# Screening of Boronic Acid Compounds

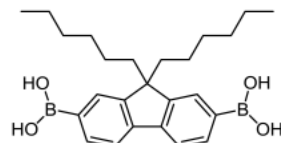
Table 1. Screening of boronic acid compounds

Entry	Boronic acid	Structure
BA01	3-Aminophenylboronic acid	
BA02	4-chlorophenylboronic acid	
BA03	4-carboxyphenylboronic acid	
BA04	Naphthalene-1-boronic acid	
BA05	3-Nitrophenylboronic acid	
BA06	Benzene-1,4-diboronic acid	
BA07	2-Naphthylboronic acid	

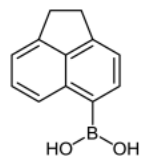
BA08	1-Thianthrenylboronic acid	
BA09	4-Cyanophenylboronic acid	
BA10	4-Methyl-1-naphthaleneboronic acid	
BA11	6-Methoxy-2-naphthaleneboronic acid	
BA12	6-Ethoxy-2-naphthaleneboronic acid	
BA13	3-Biphenylboronic acid	
BA14	8-Quinolinyboronic acid	
BA15	Pyrene-1-boronic acid	

# Screening of Boronic Acid Compounds

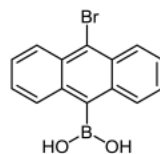
BA16 9,9-Dihexylfluorene-2,7-diboronic acid



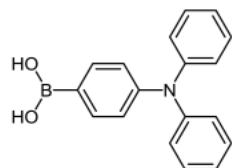
BA17 Acenaphthene-5-boronic acid



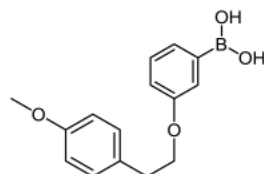
BA18 10-Bromoanthracene-9-boronic acid



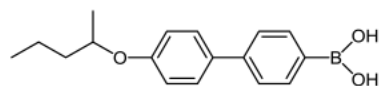
BA19 4-(Diphenylamino)phenylboronic acid



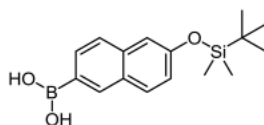
BA20 4-(4'-Methoxybenzyloxy)phenylboronic acid



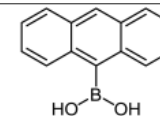
BA21 4-(4'-(2-Pentyloxy)phenyl)phenylboronic acid



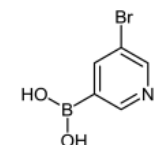
BA22 2-(tert-Butyldimethylsilyloxy)naphthalene-6-boronic acid



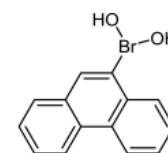
BA23 9-Anthraceneboronic acid



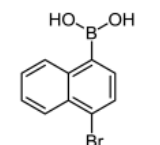
BA24 5-Bromopyridine-3-boronic acid



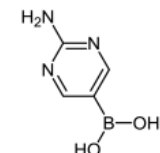
BA25 9-Phenanthracenylboronic acid



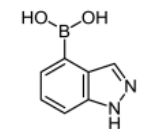
BA26 4-Bromo-1-naphthaleneboronic acid



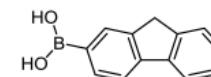
BA27 2-Aminopyrimidine-5-boronic acid



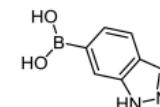
BA28 Indazole-4-boronic acid hydrochloride



BA29 Fluorene-2-boronic acid

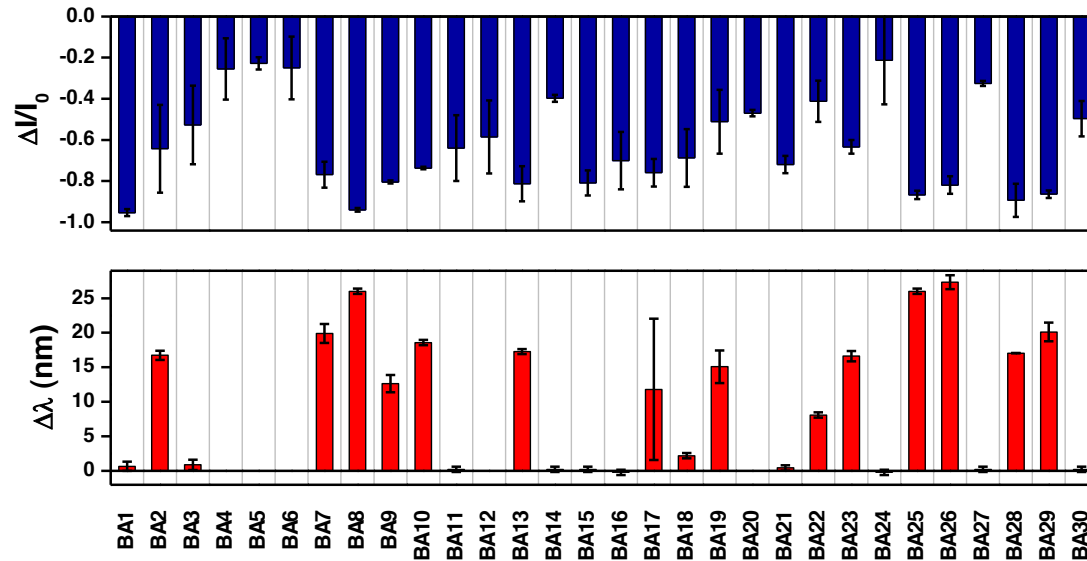


BA30 Indazole-6-boronic acid

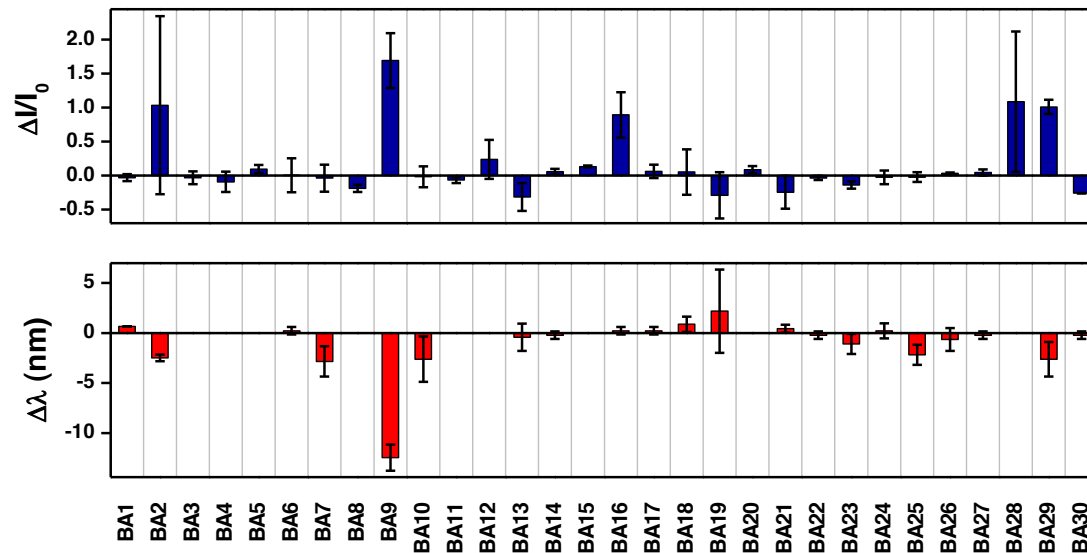


# Screening Results

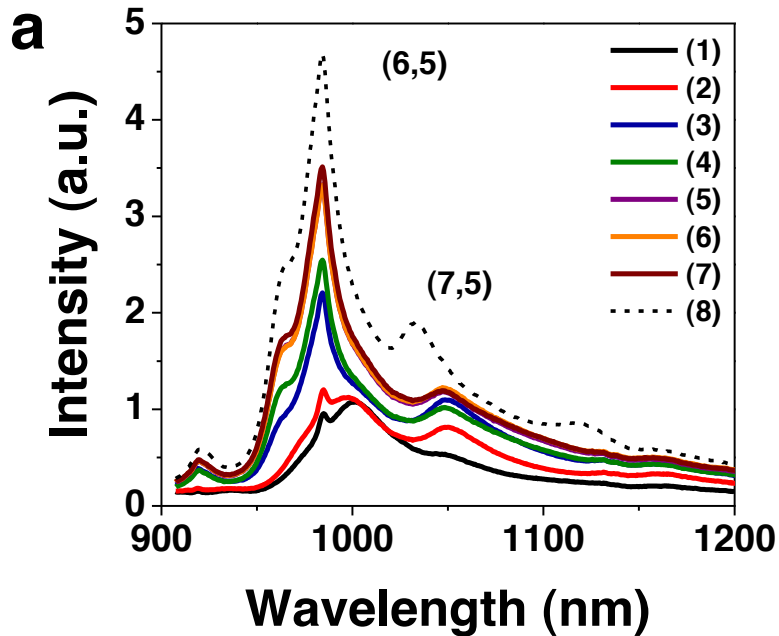
## 1. Addition of BA to SC/SWNT



## 2. Addition of glucose to BA + SC/SWNT



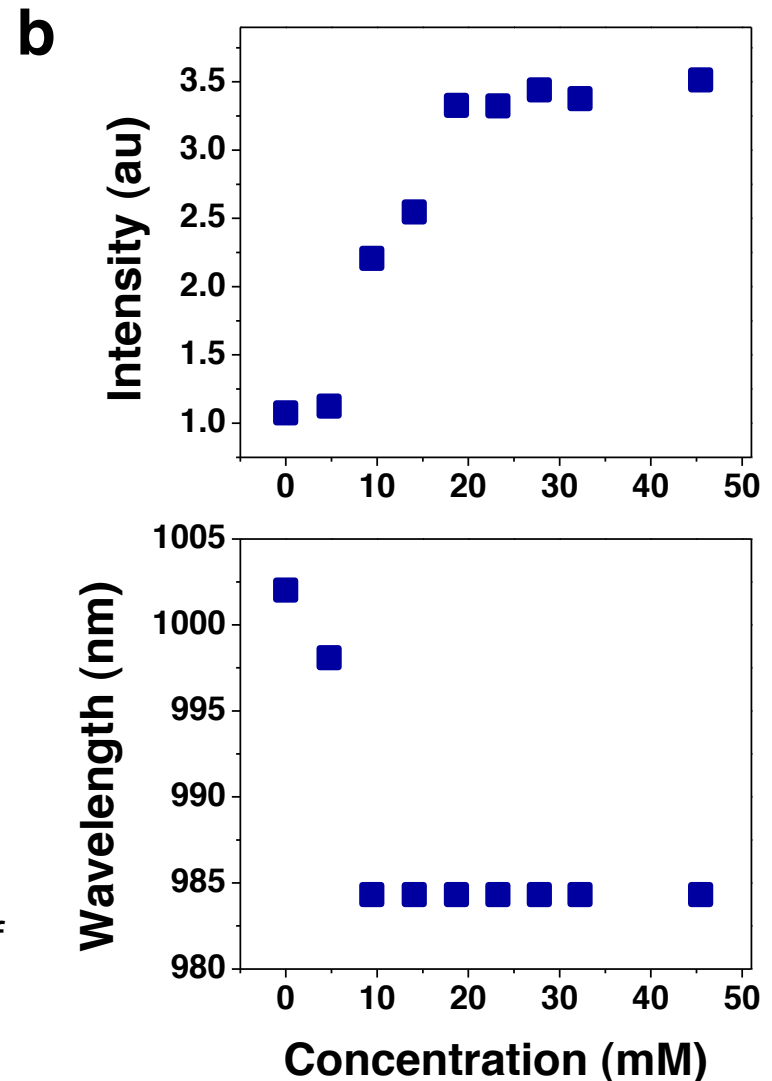
# Concentration Curve of Glucose Response (BA9)



**Fluorescence recovery of the quenched BA-SWNT complex in the presence of glucose:**

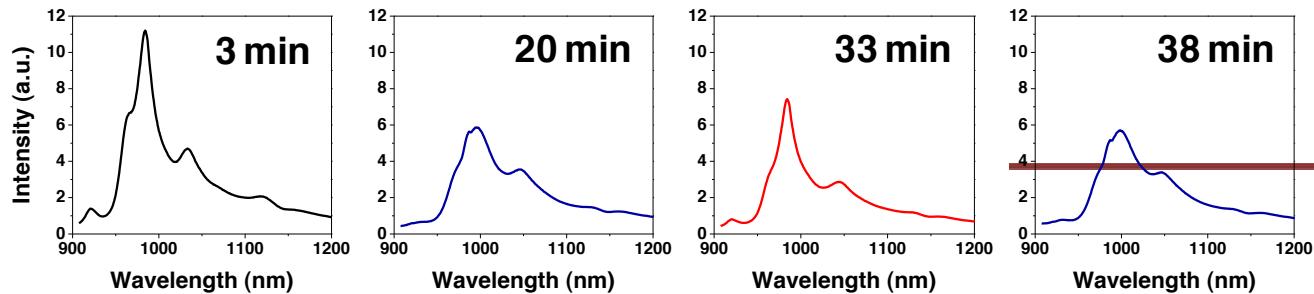
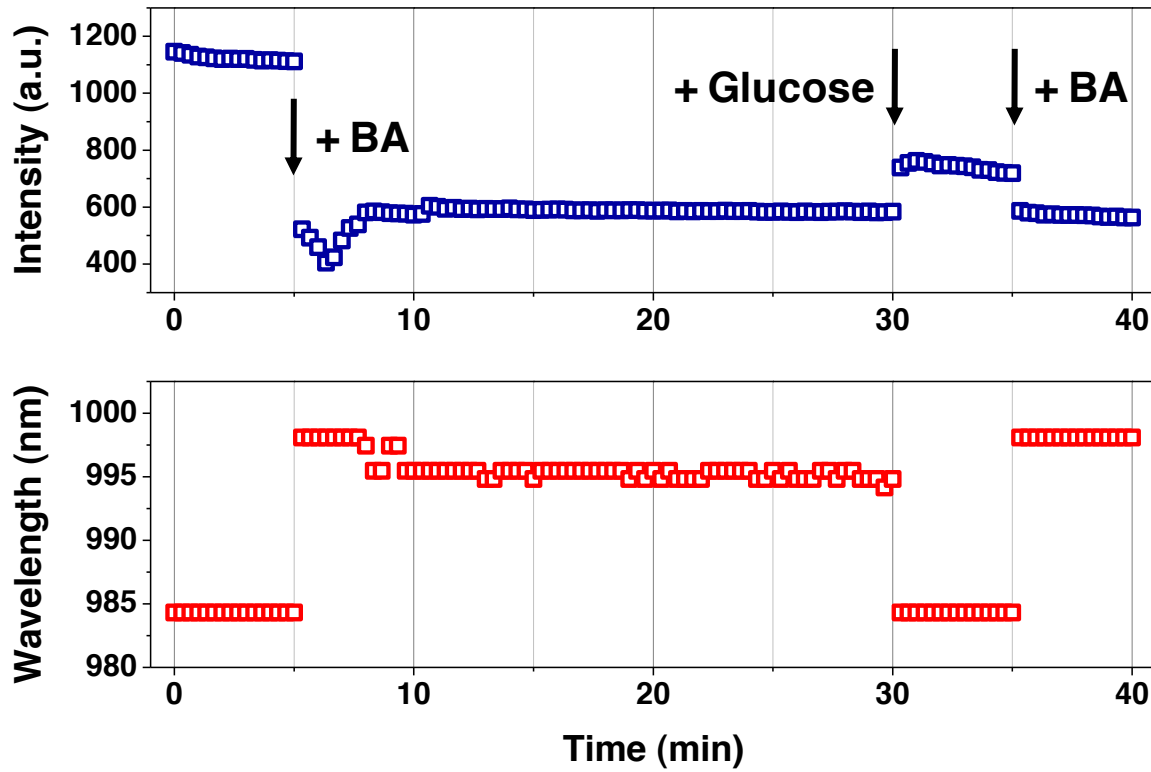
(1) Starting BA-SWNT complex of 4-cyanophenylboronic acid (BA9) and the addition of glucose of 4.7 mM (2), 9.4 mM (3), 14.1 mM (4), 18.7 mM (5), 23.3 mM (6), and 45.5 mM (7).

\* The dashed black line (8): the original spectrum of SC/SWNTs.



Fluorescence intensity and emission wavelength as a function of glucose concentration.

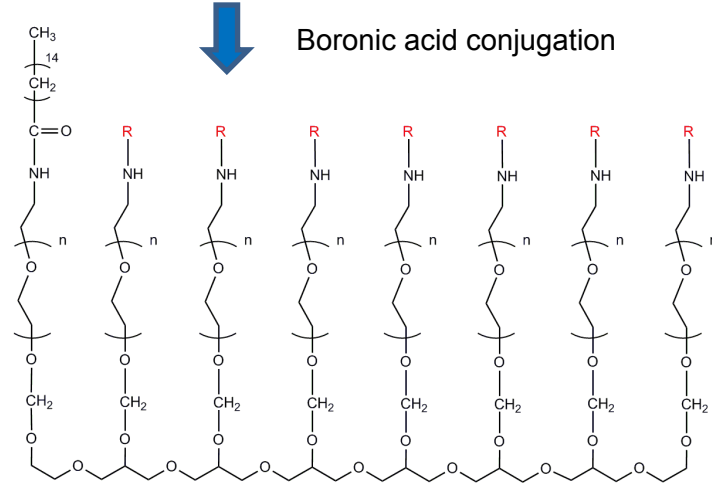
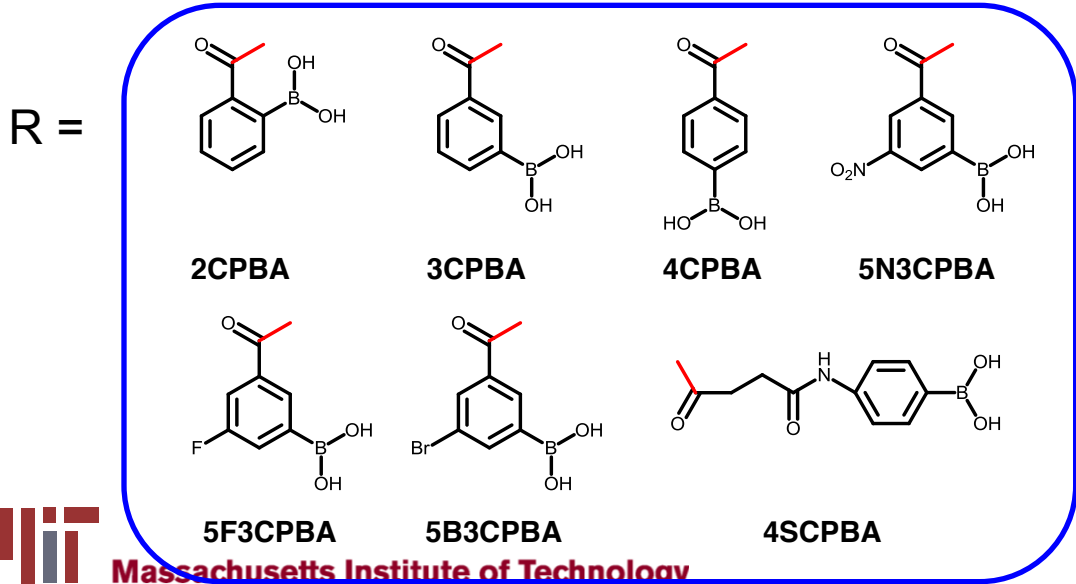
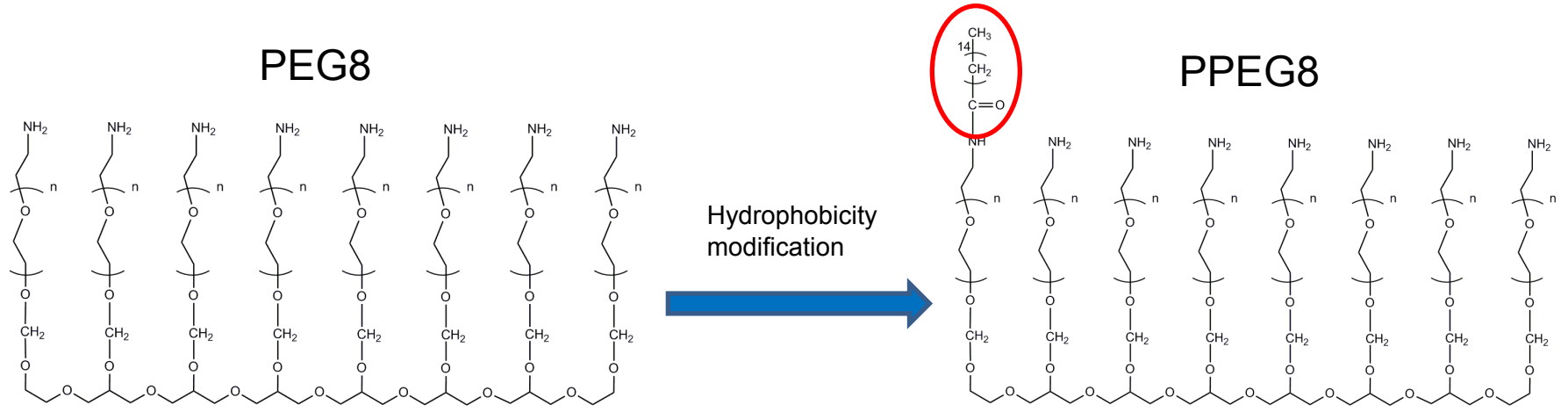
# Rapid, Reversible Dynamic Modulation



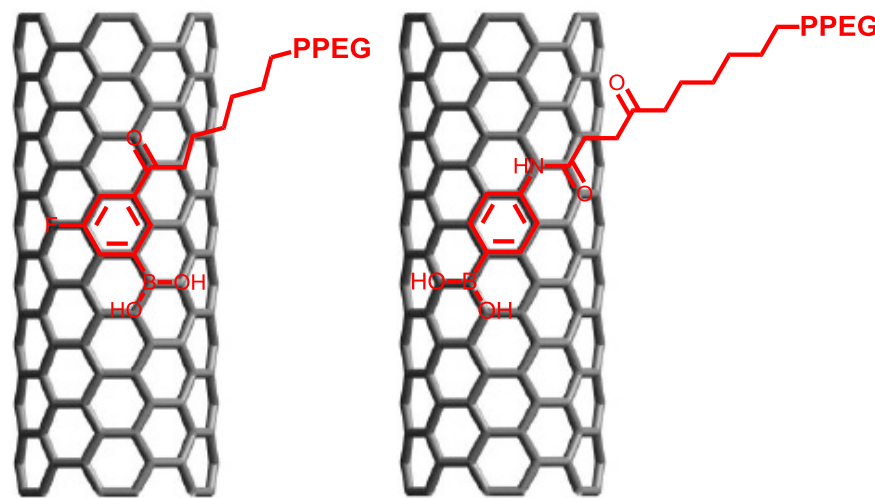
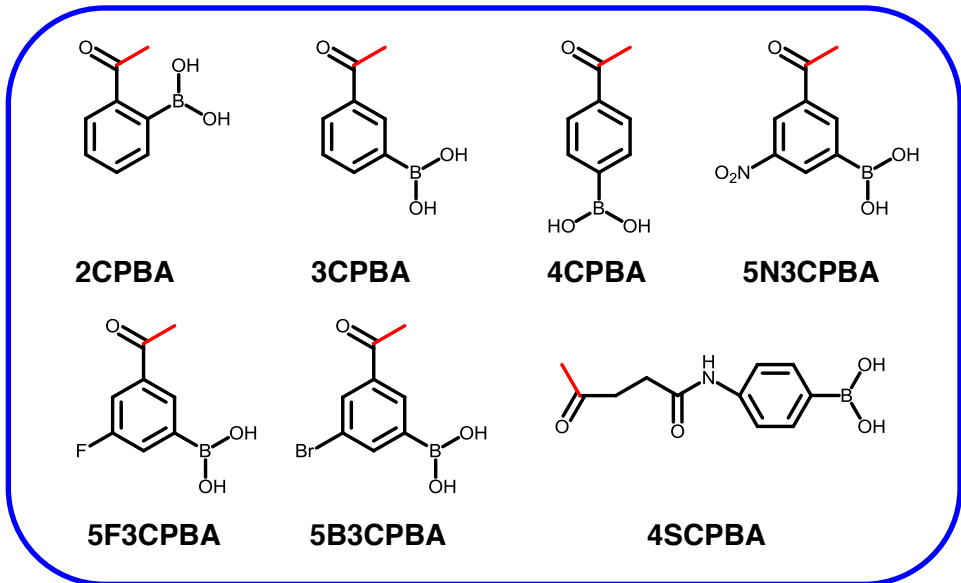
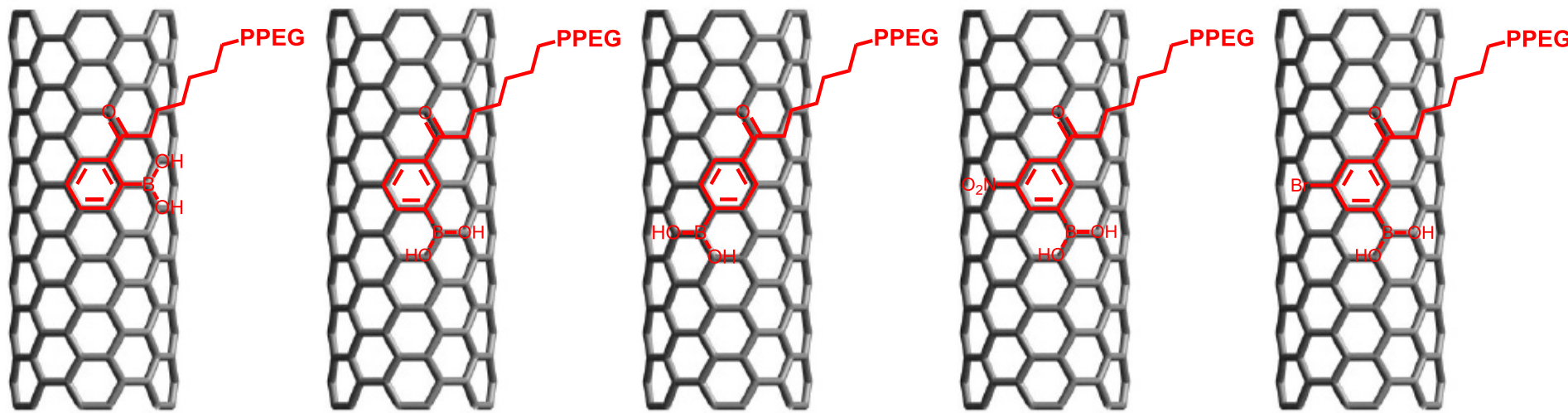
\* Fast binding and unbinding kinetics of glucose sensing

# Polymer Synthesis System: Boronic Acid Conjugated Polymers

- Molecular structure of PEG8 modified polymers



# Proposed Wrapping Geometry of Boronic Acids on SWNT

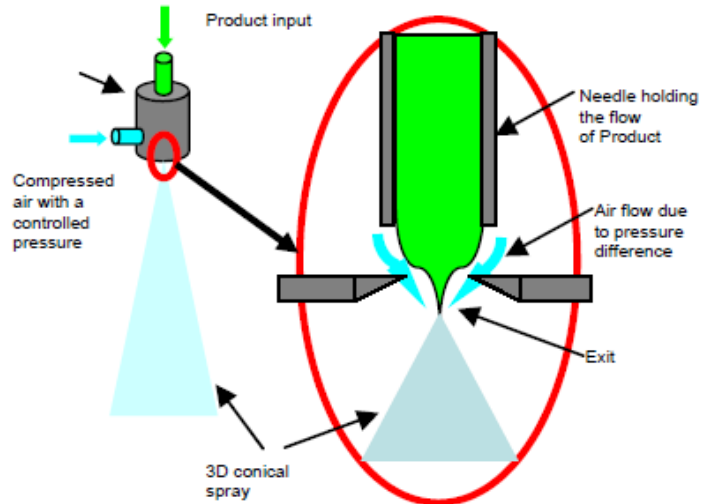


# Encapsulator for Making Hydrogel Bead

VAR J30



## Working Principle: Aerodynamic Jetting



## Flow Focusing: A Versatile Technology to Produce Size-Controlled and Specific-Morphology Microparticles\*\*

Lucía Martín-Banderas, María Flores-Mosquera, Pascual Riesco-Chueca, Alfonso Rodríguez-Gil, Ángel Cebolla, Sebastián Chávez, and Alfonso M. Gañán-Calvo\*

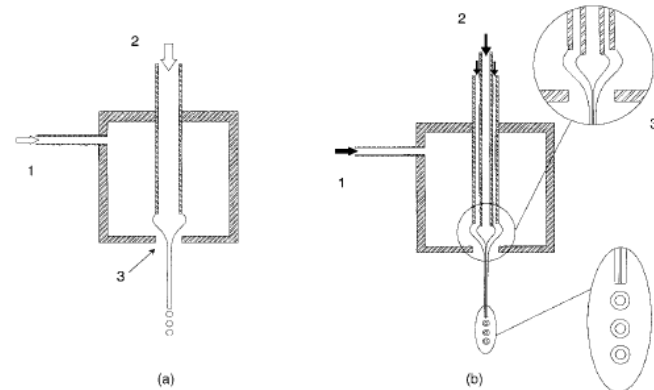


Figure 1. Flow-focusing atomizer. a) Simple jet: 1) focusing fluid, 2) focused fluid, 3) meniscus; b) compound atomizer with two concentric needles: 1) focusing fluid, 2) focused fluids: core fluid and shell fluid, 3) compound meniscus.

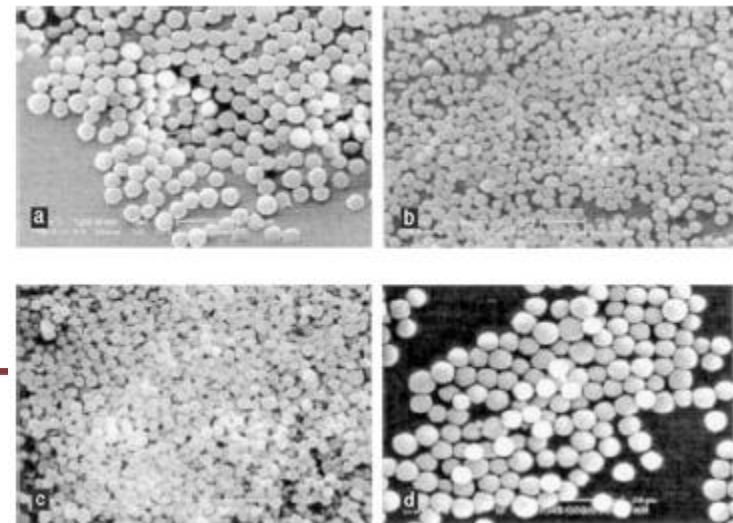
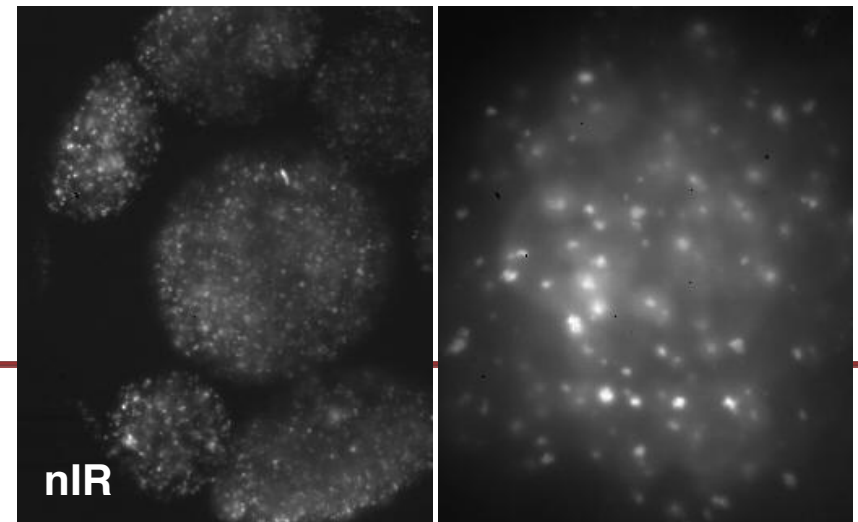
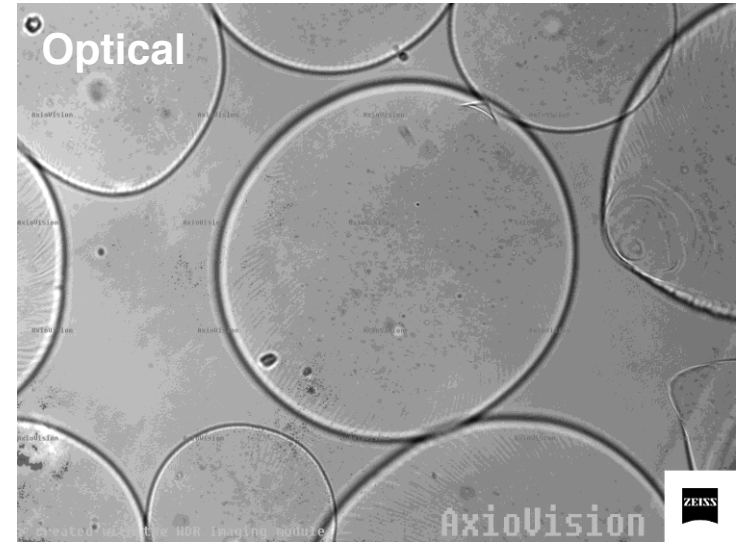
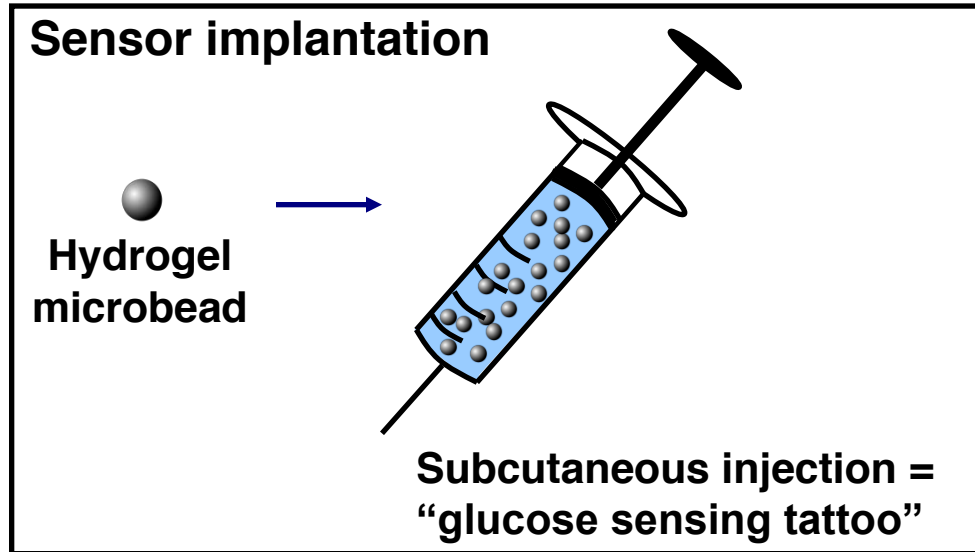


Figure 3. SEM images of freeze-dried 5 μm microparticles

# In Vivo Measurement

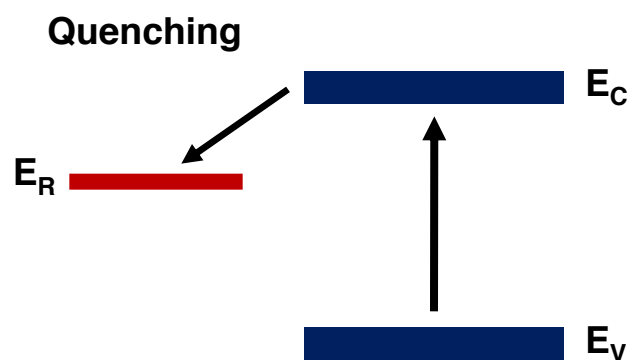
## Subcutaneous injection



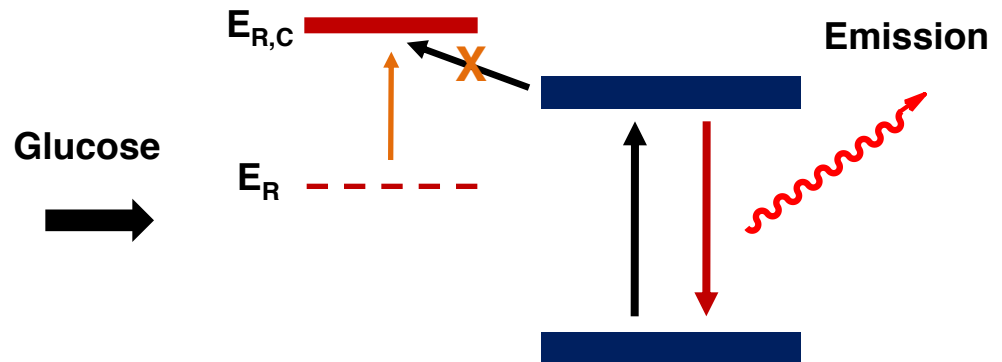
<http://web.mit.edu/newsoffice/2010/glucose-tattoo-0528.html>

# Rationale of Molecular Recognition

- 1) a change of medium polarity near SWNTs, namely solvatochromism;
- 2) electron transfer due to redox potential change;
- 3) steric effect of a quencher due to conformational, structural, or geometric change;
- 4) reversible or irreversible aggregation of SWNTs due to a change of colloidal stability.



**Fluorescence quenching mechanism:** a photo-induced excited-electron transfer



**Fluorescence recovery mechanism:** the reduction potential shift upon complexation with glucose, making the excited-electron transfer less favorable

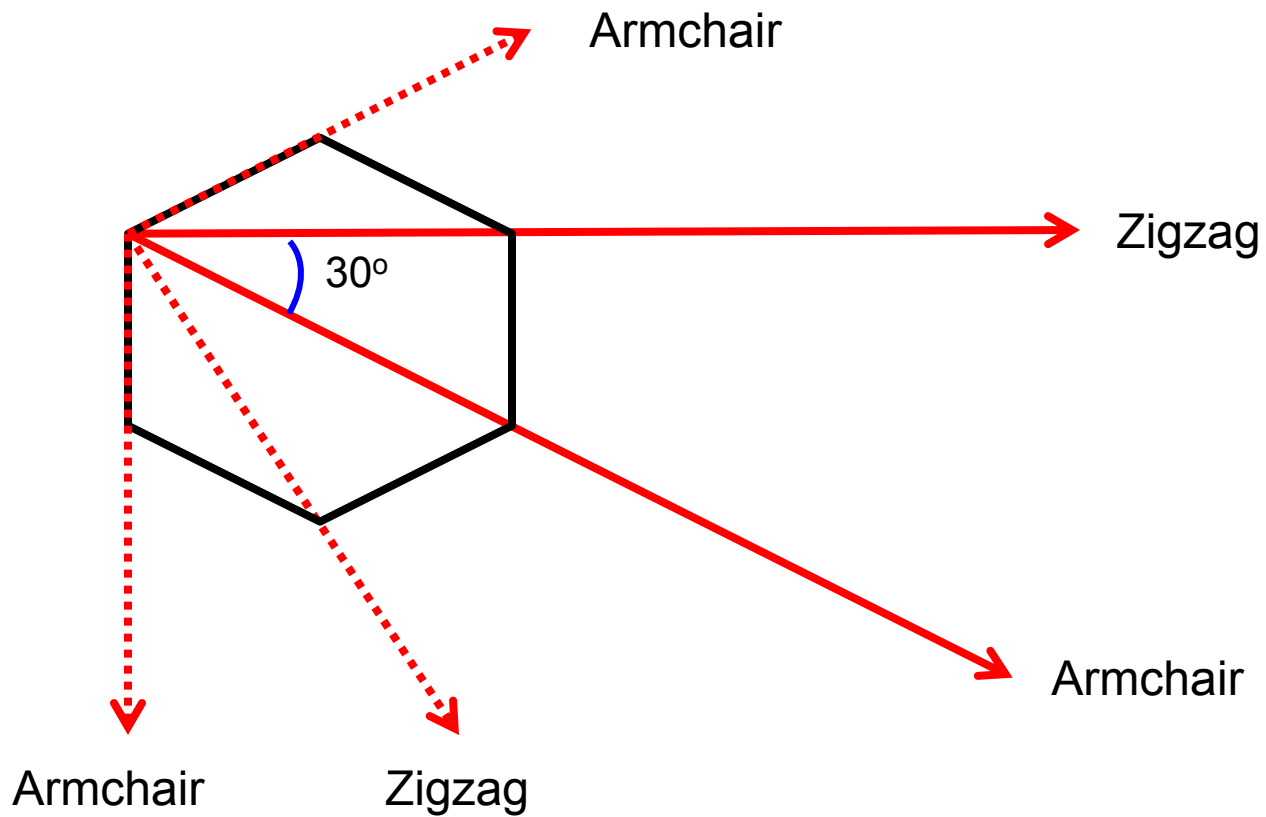
# Conclusions

- The fluorescent properties of SWNTs make them ideal for use as sensors in biological systems. Not only does the emission take place at wavelengths that do not interfere with tissue absorption, but it also demonstrates nearly infinite photostability.
- There needs to be more research into computational tools that can predict the structure of adsorbed phases around a nanotube. We need to understand how those nanoscaled structures are modified upon binding of an analyte.
- Efforts to understand interparticle potentials and solubility of SWNTs will advance their development as fluorescent sensors.

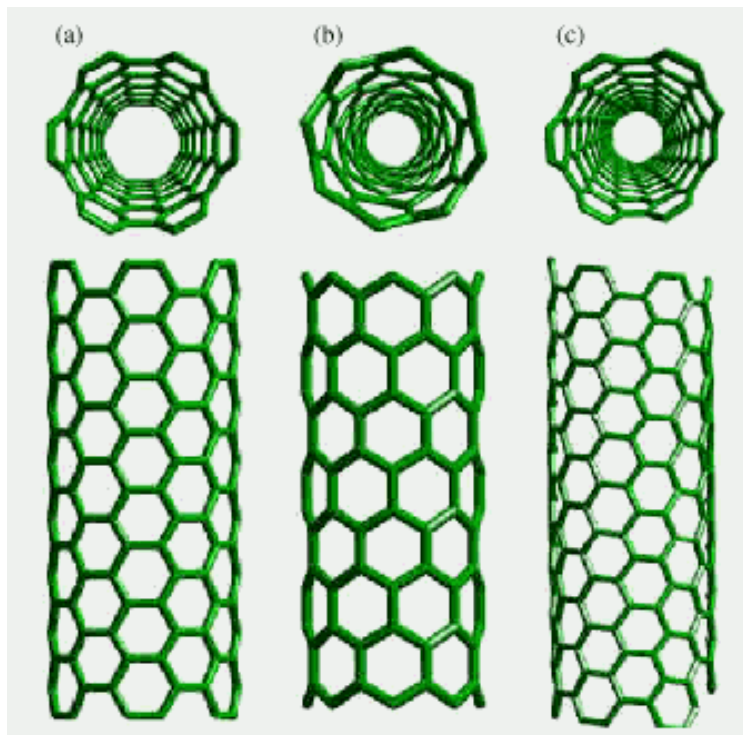
# Acknowledgement



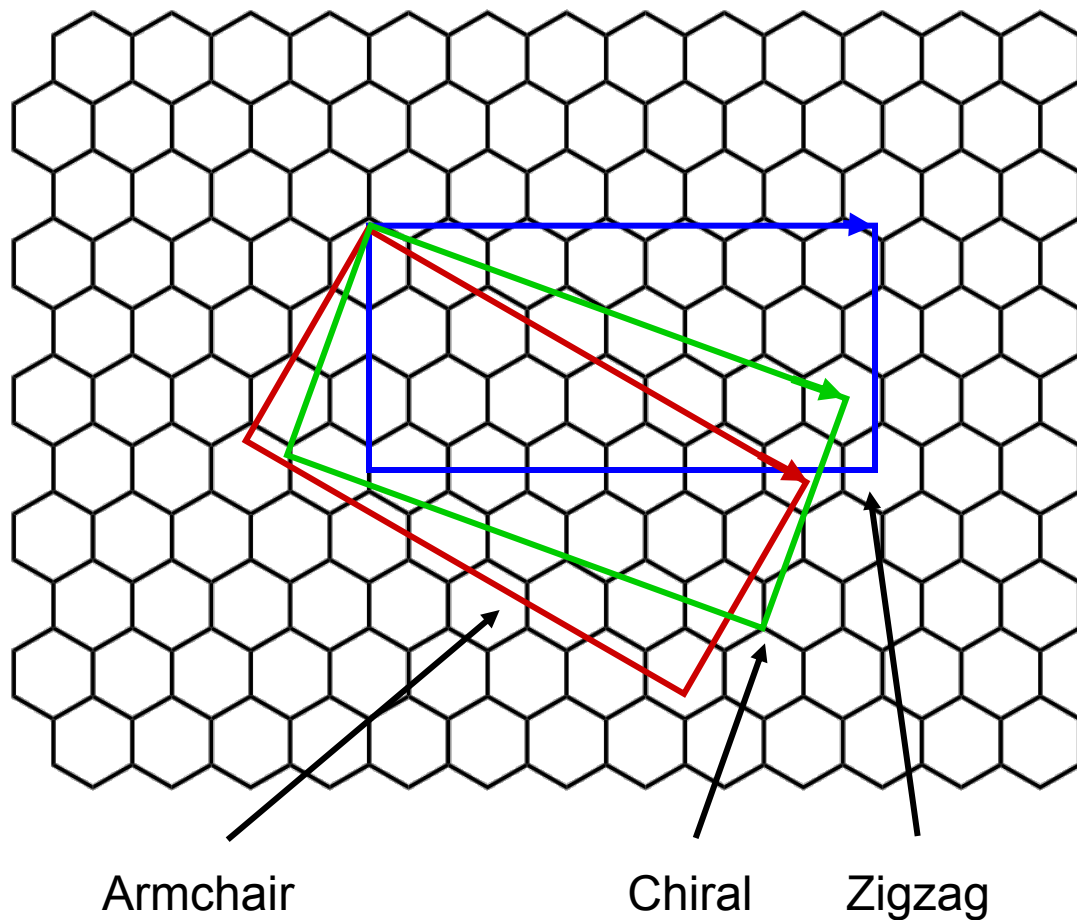
Thank You!



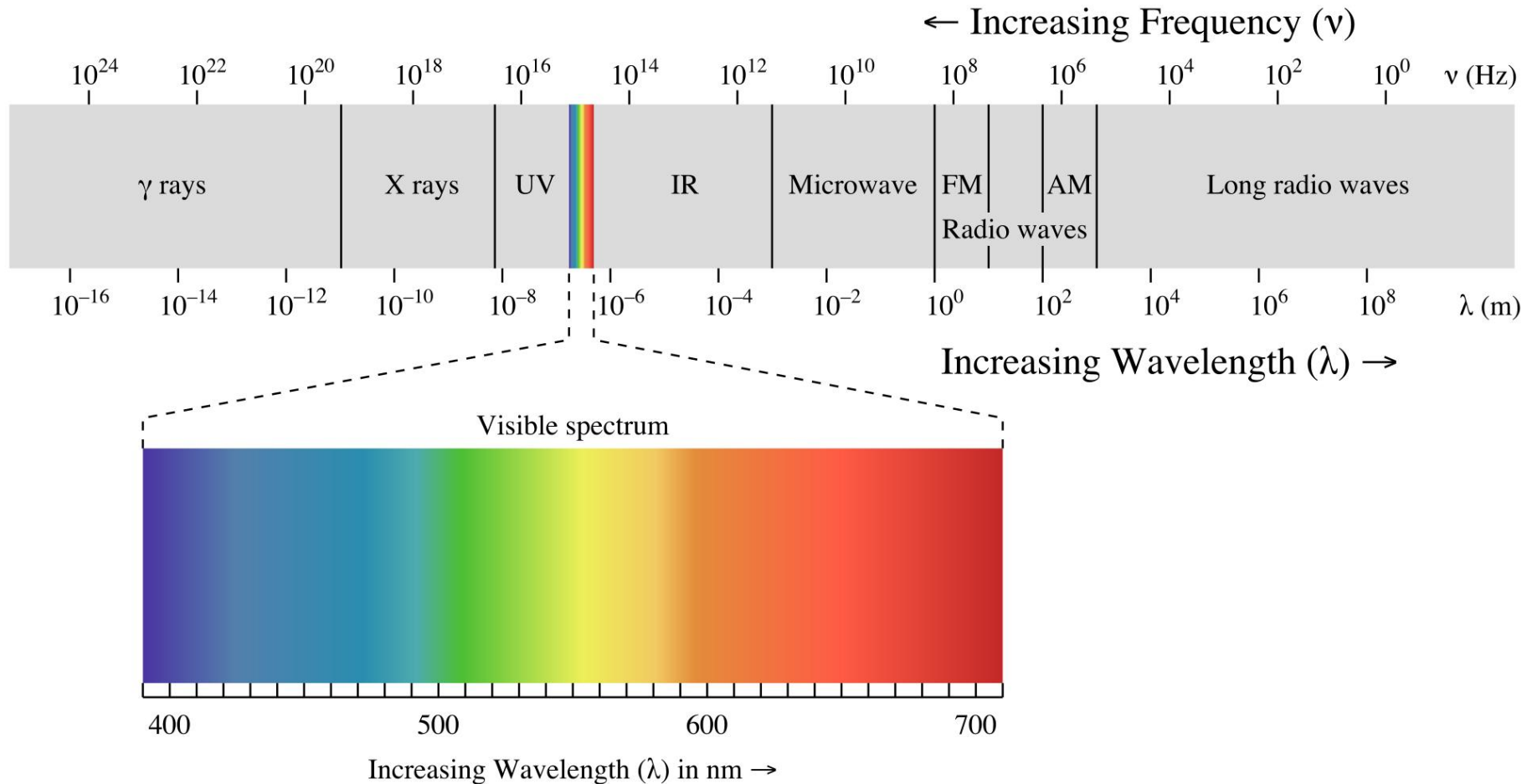
# Construction of SWNTs from a Graphene Sheet



(a) armchair, (b) zigzag, (c) chiral



# Electromagnetic Spectrum



[http://upload.wikimedia.org/wikipedia/commons/thumb/f/f1/EM\\_spectrum.svg/2000px-EM\\_spectrum.svg.png](http://upload.wikimedia.org/wikipedia/commons/thumb/f/f1/EM_spectrum.svg/2000px-EM_spectrum.svg.png)

